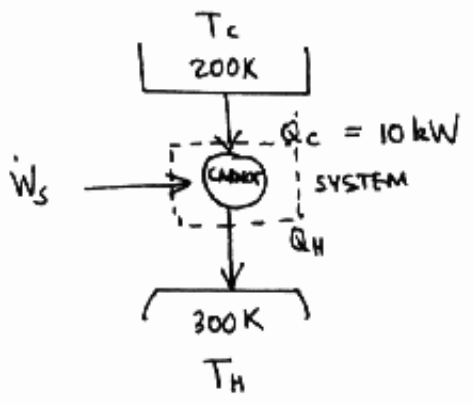


Problem Set #14

Solution

Carnot cycle



a) In this configuration, the Carnot cycle represents a HEAT PUMP

From S+VN, pg 296:

$$\omega_{\text{CARNOT}} = \frac{T_c}{T_H - T_c} \quad (9.3)$$

$$= \frac{(200)}{(300) - (200)}$$

$$\omega = 2$$

b) From 1st law, $\Delta U = 0 = \dot{Q}_c + \dot{Q}_H + \dot{W}_s$
(with Carnot heat pump as the system)

LEAVE AS +VE,
EQUATION SHOULD
TAKE CARE OF THE
SIGN.

thus, $\dot{Q}_H = -\dot{W}_s - \dot{Q}_c$
 $\dot{Q}_H = -\dot{W}_s - (10 \text{ kW}) \quad (1)$

From the definition of COEFFICIENT OF PERFORMANCE, ω :
(S+VN pg 296)

$$\omega = \frac{\text{heat absorbed at } T_c}{\text{net work}}$$

$$= \frac{|\dot{Q}_c|}{|\dot{W}_s|} \quad (9.2)$$

thus, $|\dot{W}_s| = |\dot{Q}_c| / \omega = (10 \text{ kW}) / 2 = 5 \text{ kW}$

$$\dot{W}_s = 5 \text{ kW}$$

$\dot{W}_s > 0$ since work being done on the system.

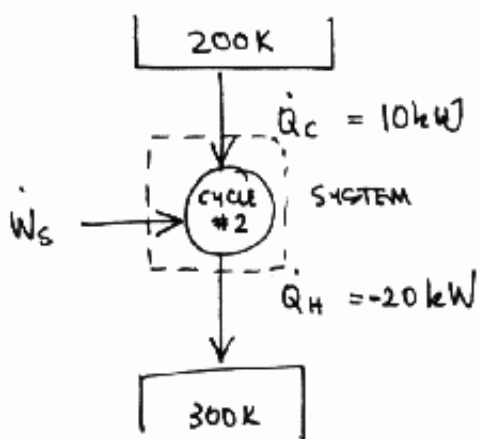
c) \diamond From eqn. (1) in part (c),

$$\begin{aligned} (1): \quad \dot{Q}_H &= -\dot{W}_S - \dot{Q}_C \\ &= -(5) - (10) \\ &= \underline{-15 \text{ kW}} \end{aligned}$$

\diamond Since \dot{Q}_H leaves system,
 $\dot{Q}_H < 0$

$$\dot{Q}_H = -15 \text{ kW}$$

Cycle # 2



d) \diamond 1st law energy balance:

$$\begin{aligned} \Delta U &= 0 \\ &= \dot{Q}_C + \dot{Q}_H + \dot{W}_S \end{aligned}$$

\diamond Thus,

$$\begin{aligned} \dot{W}_S &= -\dot{Q}_C - \dot{Q}_H \\ &= -(10) - (-20) \end{aligned}$$

$$\dot{W}_S = 10 \text{ kW}$$

(is requires 2x as much as Carnot cycle)

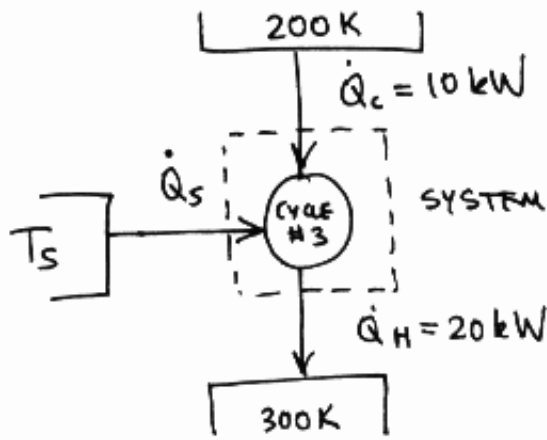
e) \diamond Coefficient of performance:

$$\omega = \frac{|\dot{Q}_C|}{\dot{W}_S} = \frac{(10 \text{ kW})}{(10 \text{ kW})} = 1$$

$$\omega_{\text{Cycle #2}} = 1$$

(lower than Carnot)

cycle # 3



f) \diamond 1st law energy balance

$$\Delta U = 0$$

$$= \dot{Q}_c + \dot{Q}_H + \dot{Q}_S$$

$$\text{so, } \dot{Q}_S = -\dot{Q}_c - \dot{Q}_H$$

$$= -(10) - (-20)$$

$$\boxed{\dot{Q}_S = 10 \text{ kW}}$$

> 0 so add to system

g) \diamond From second law, entropy balance

$$\Delta \dot{S} = 0 \quad (\text{state function like } U)$$

$$= \frac{\dot{Q}_c}{T_c} + \frac{\dot{Q}_H}{T_H} + \frac{\dot{Q}_S}{T_S}$$

$$\text{so, } \left(\frac{10 \text{ kW}}{200 \text{ K}} \right) + \left(\frac{-20 \text{ kW}}{300 \text{ K}} \right) + \left(\frac{10 \text{ kW}}{T_S} \right) = 0$$

$$\frac{10 \text{ kW}}{T_S} = -(0.05) + (0.06667) \text{ [kW]}$$

$$= 0.016667$$

$$\text{so, } T_S = \frac{(10)}{(0.016667)}$$

$$\boxed{T_S = 600 \text{ K}}$$