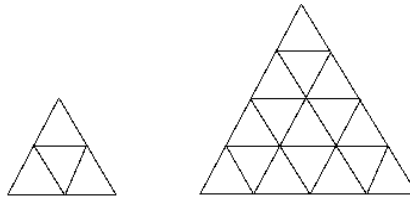


**10.420 / 10.520**  
**Problem Set #4**

1. Consider a molecule with a geometrical shape approximating a flat triangle.
- a) Determine the number of actual and potential favorable packing forces for the illustrated 4-mer and 16-mer. Assume that each edge-wise interaction with another molecule is worth  $\alpha kT/3$  per molecule.



- b) Provide the equations that generalize these forces for an n-mer.
- c) For the n-mer, derive an expression for  $\mu_0^N$  that is a function of  $\mu_0^\infty$  and N.
2. Consider a molecule that "self-assembles" into a rod at 25 °C with an interaction energy between subunits of  $\alpha kT = 24.75$  kJ/mol (10 kT).
- a) Produce plots of  $X_N$  vs. N for total concentrations, C, of 0.01 mM, 1 mM, and 10 mM for N = 1 to N = 25. (hint: you will need eqns 16.18 and 16.19 in your Israelachvili handout and/or lecture notes).
- b) Determine the value of N where  $X_N$  reaches a maximum for total concentrations, C, of 0.1 mM, 1 mM, 10 mM, 100 mM, and 1 M.
- c)  $N_{\max}$ , the value of N where  $X_N$  reaches a maximum, is a function of the concentration of C. Plot  $\log(N_{\max})$  vs.  $\log(C)$  and determine the relationship between these two parameters. In your analysis, plot all values, but only use values of  $N_{\max} > 3$
3. Provide a physical (P) and/or molecular (M) reason for the following observations. Molecular (M) rationales should include a rough figure.
- a) Hexadecane wets glass (P).
- b) Hexadecane containing stearic acid,  $\text{CH}_3(\text{CH}_2)_{16}\text{CO}_2\text{H}$ , does not wet glass (M).
- c) Addition of small amounts of sodium dodecyl sulfate (SDS) increases the spreading of water on polyethylene (P/M).
- d) Continued addition of SDS to water begins to have no effect on the spreading nature of water on polyethylene (M).
- e) The contact angle for a drop of rainwater on the hood of a car is greater during a falling rain than afterwards (P).

4. One method for measuring surface tension is to use a U-tube with a smaller radius on one side and a larger radius on the other. In such a system, a liquid that wets glass was determined to have a  $\Delta h$  of 19 mm between the levels of the two menisci in the U-tube.
  - a) Draw the U-tube and liquid identifying  $\Delta h$  and the two radii.
  - b) If the radii are 1 mm and 10 mm, and  $\rho$  is  $950 \text{ kg/m}^3$ , determine the surface tension of the liquid.
  
5.
  - a) Given the surface tensions of heptane ( $20.14 \text{ dyn/cm}$ ) and diethylene glycol ( $30.9 \text{ dyn/cm}$ ), calculate the works of cohesion for these solvents. The work of cohesion is the energy ( $\text{erg/cm}^2$ ) required to separate one body of liquid/material into two. If two  $1 \text{ }\mu\text{L}$  drops of heptane that are suspended in air combine to form one  $2 \text{ }\mu\text{L}$  drop, estimate the energy gain ( $\text{erg/cm}^3$ ) for this process and a temperature rise for the heptane.
  - b) Given the interfacial tension of heptane-diethylene glycol ( $10.6 \text{ dyn/cm}$ ), calculate the work of adhesion for the heptane-diethylene glycol interface.
  
6.
  - a) A fabric is made of wool fibers of individual diameter  $20 \text{ }\mu\text{m}$  and density of  $1.3 \text{ g/cm}^3$ . The contact angle for water on a single fiber is  $120^\circ$ . Calculate the contact angle of water on fabric woven so that its bulk density is  $0.8 \text{ g/cm}^3$ .
  - b) If the fibers are chemically modified so that the contact angle of water on the individual fiber is  $60^\circ$ , what would be the contact angle on the above woven fabric?