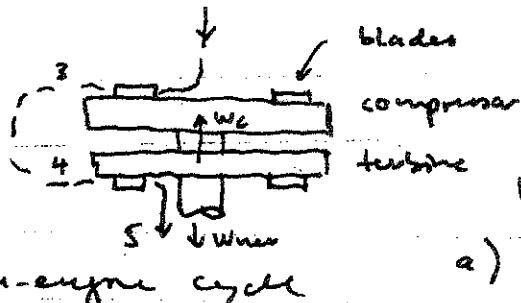


T5



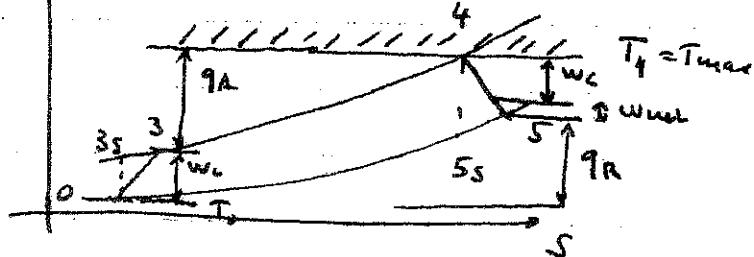
16. Modified Sp 09

25

(all states are stagnation states)

a)

$$\Pi = 3, T_{max} = \frac{T_4}{T_0} = 5, \gamma_T = 0.7$$



Concepts: 1st and 2nd law, definition of thermal eff., adiab. eff.

$$b) \eta_{th} = \frac{w_{wheel}}{q_A} = 1 - \frac{q_A}{q_A} = 1 - \frac{T_5 - T_0}{T_4 - T_3} \quad \text{define } \tilde{\gamma}_T = \Pi^{\frac{1}{\delta-1}}$$

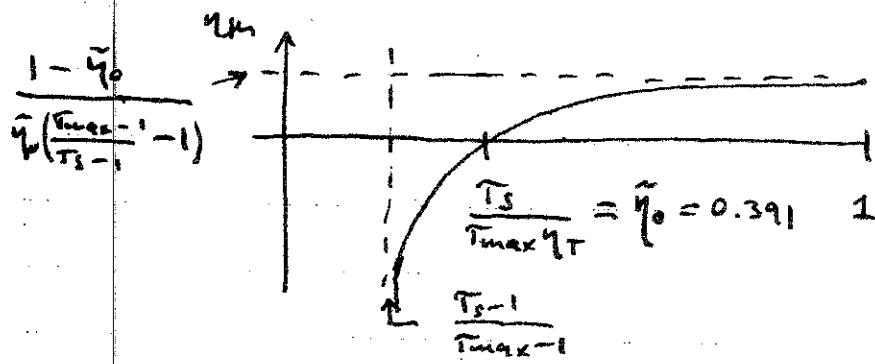
$$\text{Using def. of } \eta_T \text{ and isentropic relations: } T_5 = T_4 \left(1 - \eta_T \left(1 - \frac{1}{\tilde{\gamma}_T}\right)\right)$$

$$\text{Similarly for compressor: } T_3 = T_0 \left(1 + \frac{1}{\gamma_C} (\tilde{\gamma}_s - 1)\right)$$

$$\text{Substitute in above: } \eta_{th} = 1 - \frac{T_4/T_0 \left(1 - \eta_T \left(1 - \frac{1}{\tilde{\gamma}_T}\right)\right) - 1}{T_4/T_0 - \left(1 + \frac{1}{\gamma_C} (\tilde{\gamma}_s - 1)\right)}$$

$$\text{Using } T_{max} = T_4/T_0 \text{ find with } \tilde{\eta}_0 = \tilde{\gamma}_s / (\tilde{\gamma}_s - 1)$$

$$\eta_{th} = \frac{\eta_C [T_{max} \eta_T (1 - \tilde{\gamma}_T^{-1})] - (\tilde{\gamma}_s - 1)}{\eta_C [T_{max} - 1] - (\tilde{\gamma}_s - 1)} = \frac{\eta_C / \tilde{\eta}_0 - 1}{\eta_C \frac{T_{max} - 1}{\tilde{\gamma}_s - 1} - 1}$$



c) If $\eta_{th} > 0$ will produce net work so

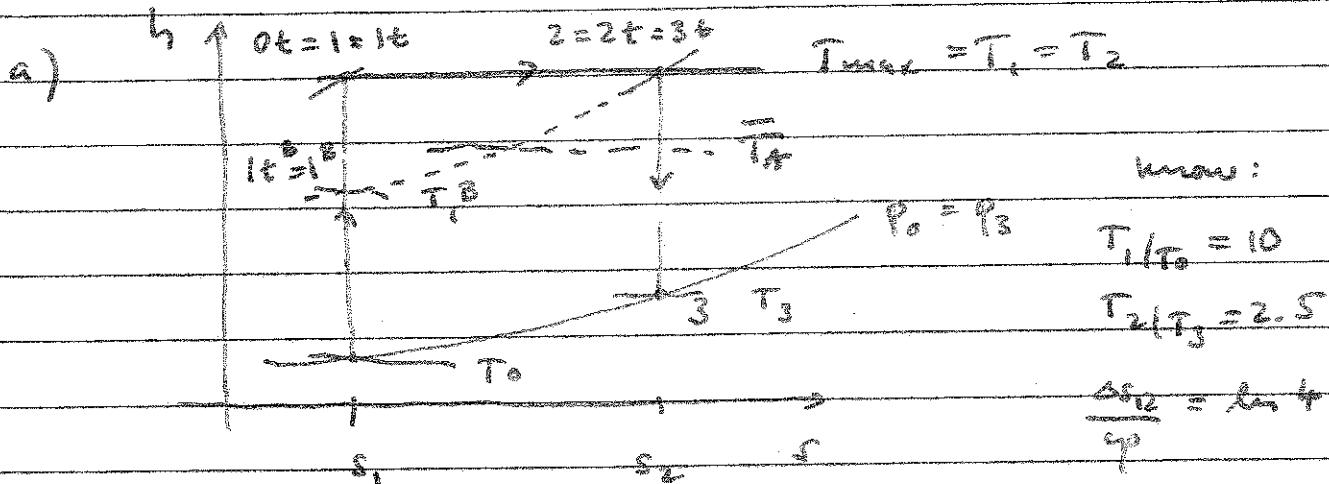
$$w_{wheel} > 0 \text{ for } \gamma_C > \frac{\tilde{\gamma}_s}{T_{max} \eta_T}$$

$$\gamma_C > 0.391$$

$$d) P = w_{wheel}, w_{wheel} = cp T_4 \eta_T (1 - \tilde{\gamma}_T^{-1}) - \frac{cp T_0}{\gamma_C} (\tilde{\gamma}_s - 1), \text{ for } \gamma_C = 0.55 \rightarrow w_{wheel} = 82 \frac{W}{kg}$$

$$w_{wheel} = g \rho_{air} A_{exit} V_{exit}; P_{exit} = P_0 \left(1 + \frac{k-1}{2} M_{exit}^2\right)^{\frac{-k}{k-1}}, T_{exit} = T_0 \left(1 + \frac{k-1}{2} M_{exit}^2\right)^{-1}$$

$$\dot{m}_{exit} = \frac{P_{exit}}{R T_{exit}}, V_{exit} = M_{exit} \sqrt{RT_{exit}}, \rightarrow \text{find } P = 32.9 \text{ W}$$



$$5) \quad q_{\text{eff}} = \frac{w_{\text{ext}}}{q_A} = 1 + \frac{q_B}{q_A} \quad ; \quad \rightarrow 2: \quad Tdq = dq \quad | \cdot \int$$

(July 1860)

$$320 : Td = \Delta q = \Delta h$$

$$\frac{d}{dt} \ln \left(\frac{T_0 - T}{T_0} \right) = - \frac{\Delta S_{\text{m}}}{\Delta H_{\text{m}}} \frac{dT}{T}$$

$$q_{1R} = q_2(T_0 - T_1) < 0$$

$$\gamma_{\text{H}_2} = 1 - \frac{T_2(\tau_1 - \tau_2/\tau_1)}{\Delta E / k_B} + \gamma_{\text{H}_2} = 1 - \frac{12.5 - 11.6}{2 \times 4} \quad \gamma_{\text{H}_2} = 0.72$$

c) heat added at const pressure, same T_{max} , T_2/T_3

$$\frac{T_{\text{R}}}{T_{\text{K}}} = \frac{T_0}{T_{\text{R}}} \quad \Rightarrow \quad \text{same PR} \rightarrow \text{same TR} : \frac{T_0}{T_0} = \frac{T_{\text{R}}}{T_{\text{R}}}$$

$$\frac{e^2}{4\pi\epsilon_0} = 0.6$$

$T_K < T_m$ because heat added at average temperature
(lower than max. cycle temp. (const. in a))