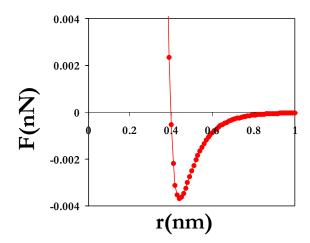
## ROBLEM SET 7 DUE FRIDAY NOVEMBER 7<sup>th</sup>

## MOHR'S CIRCLE

- 1. Given a plane with stresses  $\sigma_x$  and  $\sigma_y$  and  $\tau_{xy}$ , draw the Mohr's circle if the plane rotates 45° clockwise to the current axes. Hint:  $\tau_{xy} < 0$  and  $\sigma_x > \sigma_y$
- 2. Given a plane with stresses  $\sigma_x$  and  $\sigma_y$  and  $\tau_{xy}$ , draw the Mohr's circle if the plane rotates counterclockwise to the current axes between 45-90degrees. Hint:  $\tau_{xy} > 0$  and  $\sigma_x < \sigma_y$
- 3. Given a plane with stresses  $\sigma_x$  and  $\sigma_y$  and  $\tau_{xy}$ , draw the Mohr's circle if the principle axes are clockwise to the current axes and between 45-90 degrees away. Hint:  $\tau_{xy} < 0$  and  $\sigma_x < \sigma_y$
- 4. Given a plane with stresses  $\sigma_x$  and  $\sigma_y$  and  $\tau_{xy}$ , draw the Mohr's circle if the principle axes are aligned with current axes. Hint:  $\tau_{xy} = 0$  and  $\sigma_x > \sigma_y$

## LENNARD-JONES POTENTIAL

5. Two atoms interact at T=0°K via a van der Waals Lennard-Jones potential with A=4.7•10<sup>-78</sup> Jm<sup>6</sup>. The interaction force versus separation distance plot is given in the following figure.



- (a) Calculate the binding energy ( $k_BT$ ) and the bond stiffness (N/m).
- (b) The two atoms are held at a particular separation distance r using an atomic force microscope so that the *attractive component* of the force is equal to -0.003 nN. At this distance are the atoms attracted to each other or repelled away from each other? Justify your answer with a numerical calculation.

- 6. Draw the following Common Potential Functions based on the information given. The capital Gammas represent your Potential functions U(r) that was discussed in class. (Note: some may be impossible)
- 1) Ideal Gas

$$\Gamma(r) = 0$$

2) Hard Sphere

$$\Gamma(r) = \infty (r \le \sigma)$$
  
$$\Gamma(r) = 0 (r > \sigma)$$

3) Point Repulsion

$$\Gamma(r) = dr^{-\delta}$$
  
 $\delta = \text{index of repulsion;}$   
 $9 < \delta < 15$   
If  $\delta = 4 \implies$  "Maxwellian molecules"

4) <u>Square Well</u>

$$\overline{\Gamma}(r) = \infty \quad (r \le \sigma)$$

$$\Gamma(r) = -\varepsilon \quad (\sigma < r \le R\sigma)$$

$$\Gamma(r) = 0 \quad (r > R\sigma)$$

5) Sutherland

$$\Gamma(r) = \infty \quad (r \le \sigma)$$
 $\Gamma(r) = -cr^{-\gamma} \quad (r > \sigma)$ 
Typically,  $\gamma = 6$ 

6) <u>Lennard-Jones</u>

$$\Gamma(r) = 4\varepsilon \left[ \left( \frac{\sigma}{r} \right)^{12} - \left( \frac{\sigma}{r} \right)^{6} \right]$$

## 7) <u>Buckingham</u>

$$\Gamma(r) = b \exp(-ar) - \frac{c}{r^6} - \frac{c'}{r^8}$$

- 4-parameter
- exponential form for repulsive (theoretically better)
- includes induced dipole / induced dipole
   å induced dipole / induced quadrupole
- numerically difficult