

# **Workshop on a “Drug Discovery” Approach to Breakthroughs in Batteries**

## **Introduction**

**Nick Delgass**

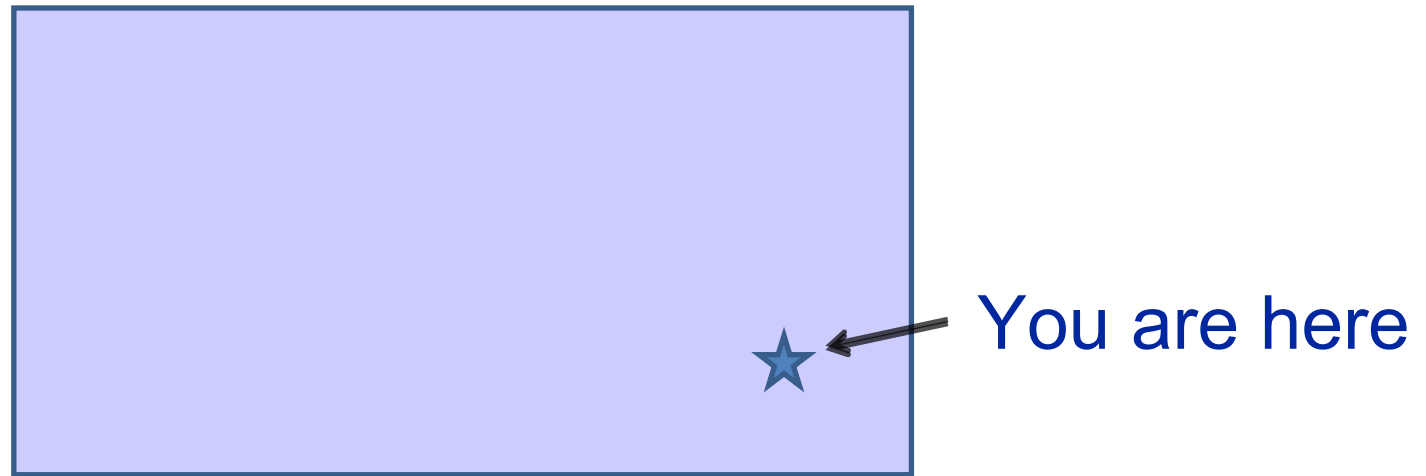
**School of Chemical Engineering**

**Purdue University**

**(Core expertise – heterogeneous catalysis)**

# The Problem with Discovery

- **NOT business as usual**
- **Descriptor (search) space is huge –  $10^6$  –  $10^{18}$**   
e.g. elements of the periodic table 4 at a time



- **NOT only traditional scientific method**
- **Breakthrough means NOT an extrapolation from where you are. Need new approach.**

# Navigating Descriptor Space

- **Designed broad search – statistical analysis**  
**Fast, helps identify descriptors**
- **Model-based search – incorporate knowledge, including fundamental**
- **“Only” need better, not best, performance**
- **Intelligent search is an absolute necessity**
- **Need a lot of data (information)**

# Navigating Descriptor Space

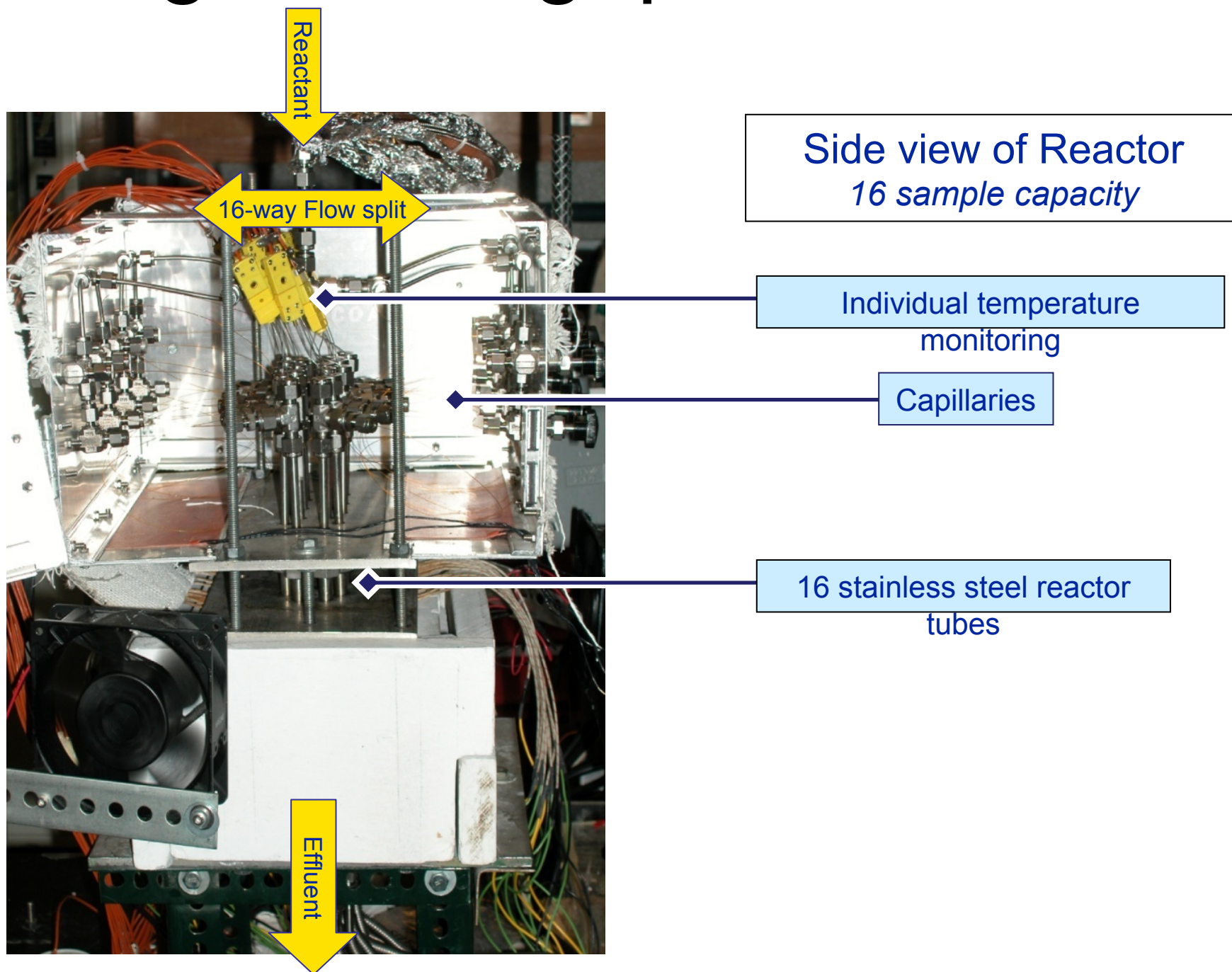
- **Experimental Data (parallel) - synthesis, performance, and materials characterization**
  - Primary* (high throughput, 1 or 2 parameters)
  - Secondary* (lower data rate, more parameters)
  - Tertiary* (lowest data rate, real conditions)
- **Theory –**
  - Guide or drive models, calculate descriptors, cover space not accessible to experiments
- **Informatics – Handling and using data**
  - Searching, sharing, and archiving
  - Tools for knowledge extraction, model building
- **TEAMS – Success requires all of the above**

# **Example**

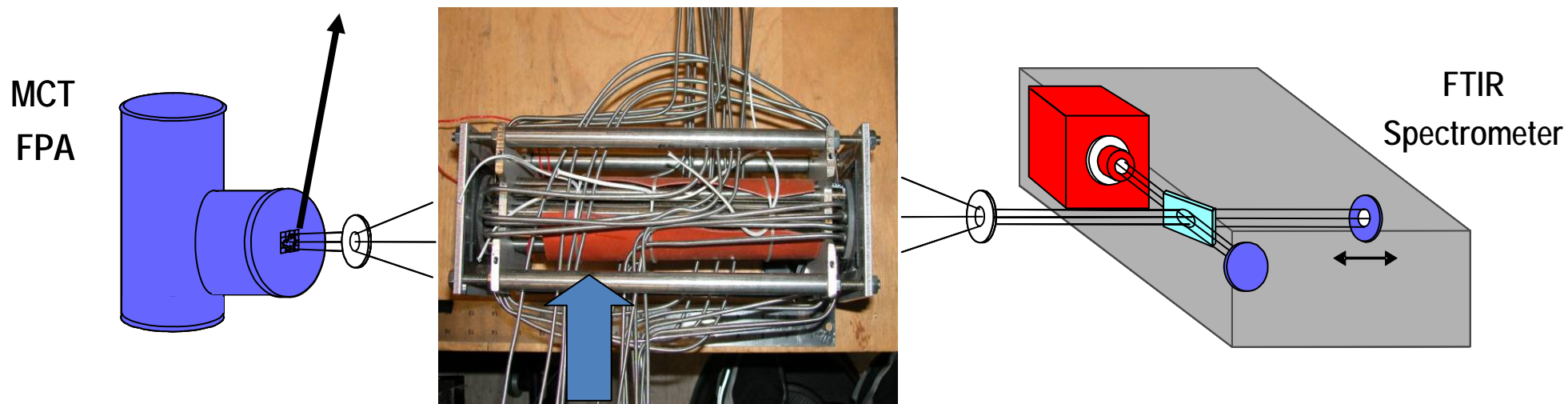
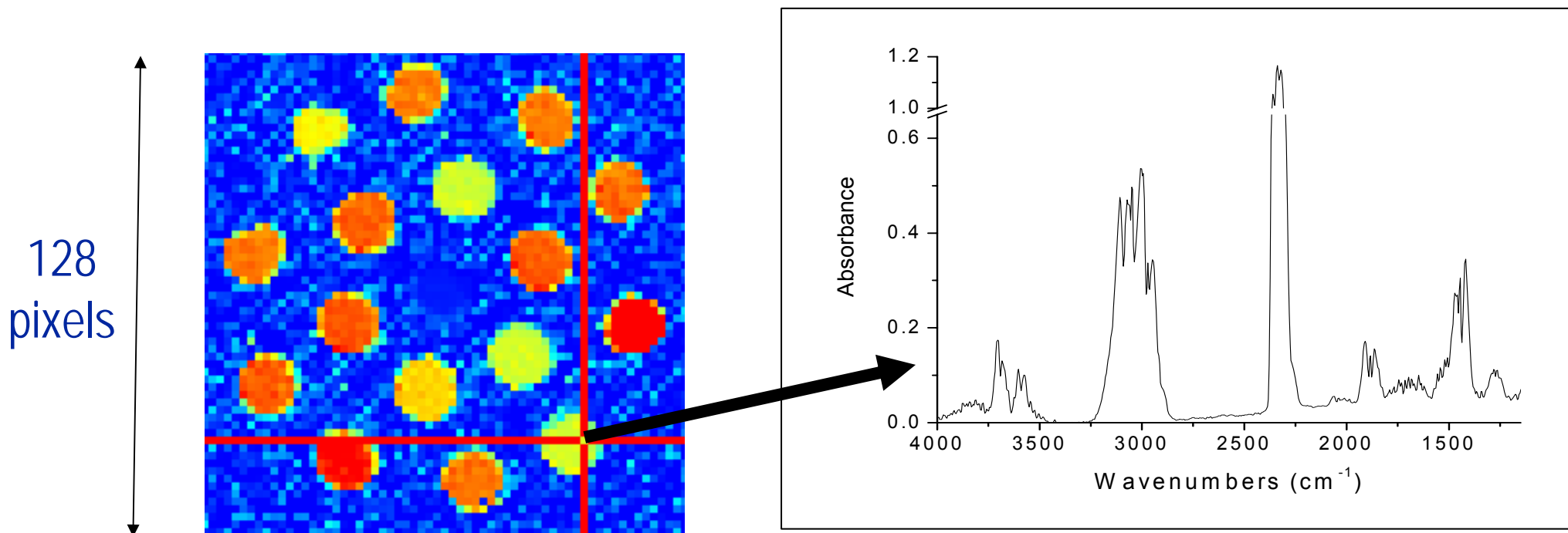
## **Search for a new catalyst**

**W. Pryz, R. Vijay, J. Binz, Jochen Lauterbach, and D.J. Buttrey, “Characterization of K-Promoted Ru Catalysts for Ammonia Decomposition Discovered Using High-Throughput Experimentation”, *Topics in Catalysis*, 2008, online**

# High-Throughput - Reactor

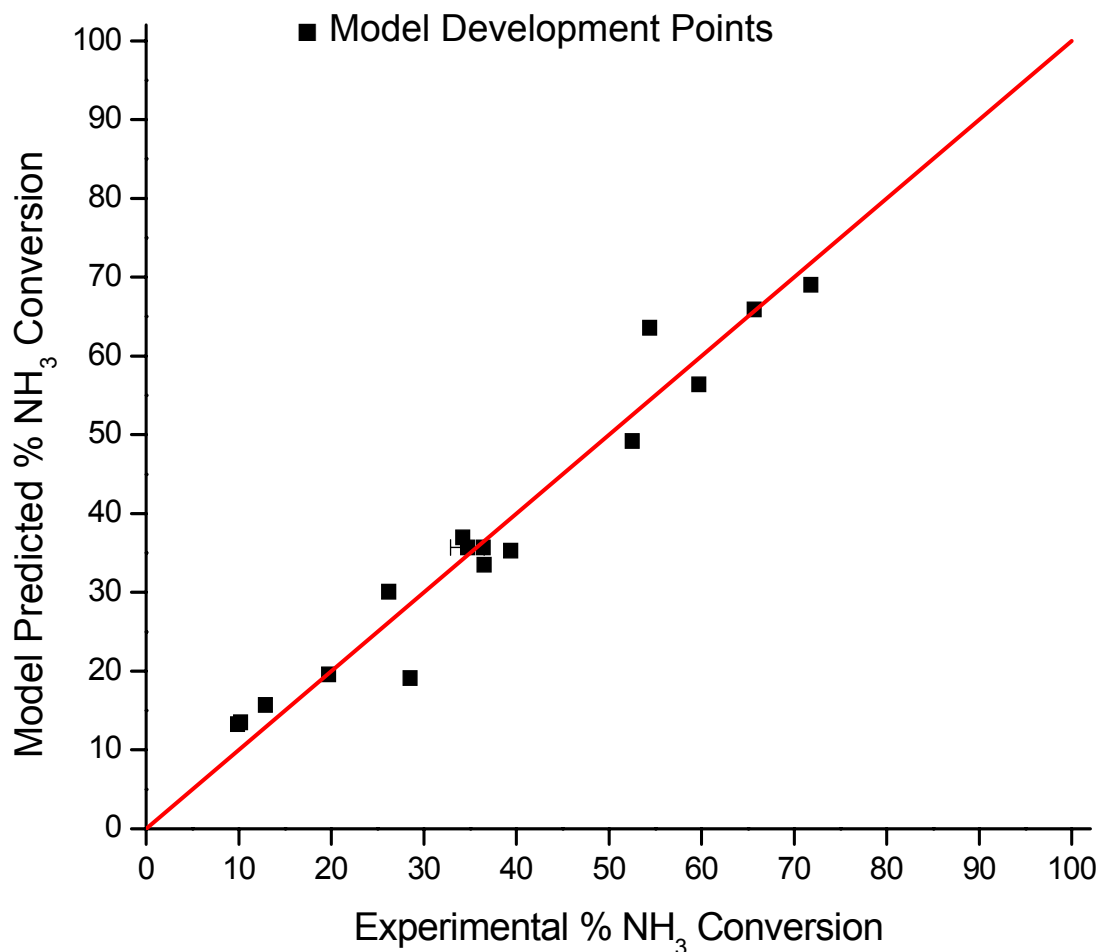
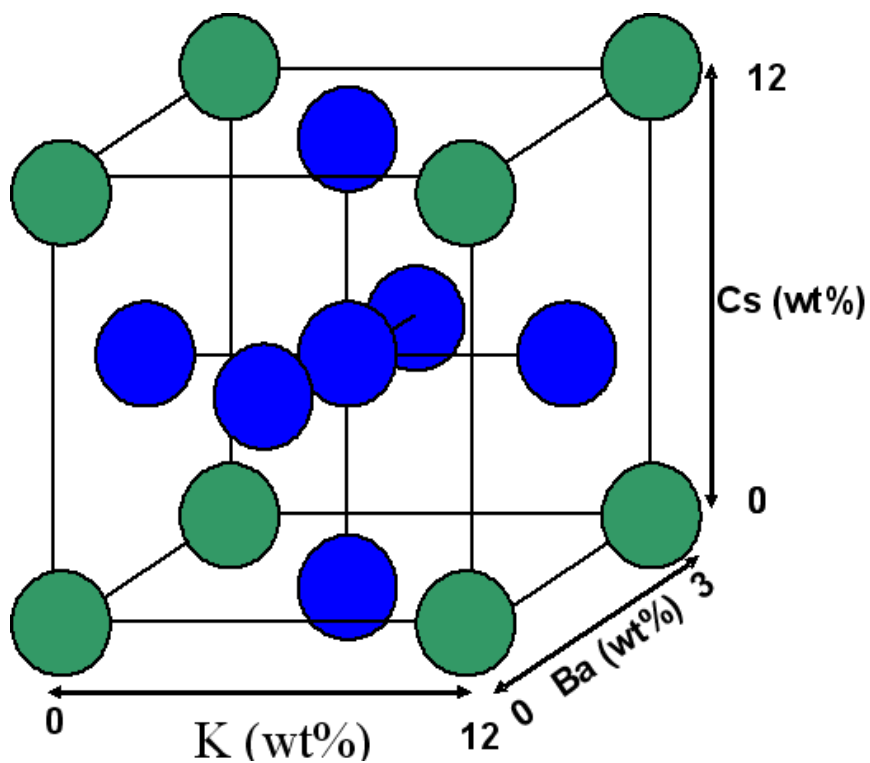


# Truly Parallel Screening



C.M. Snively, G. Oskarsdottir, and J. Lauterbach, *Angewandte Chemie*, 40(2001)  
Snively, C.M. and J. Lauterbach *Applied Spectroscopy* 59 (2005)  
R. Hendershot, P. Fanson, C.M. Snively, and J. Lauterbach, *Angewandte Chemie*, 42(10); (2003).

# The Next Step: Promotion of Ru/ $\gamma$ -Al<sub>2</sub>O<sub>3</sub> Catalysts

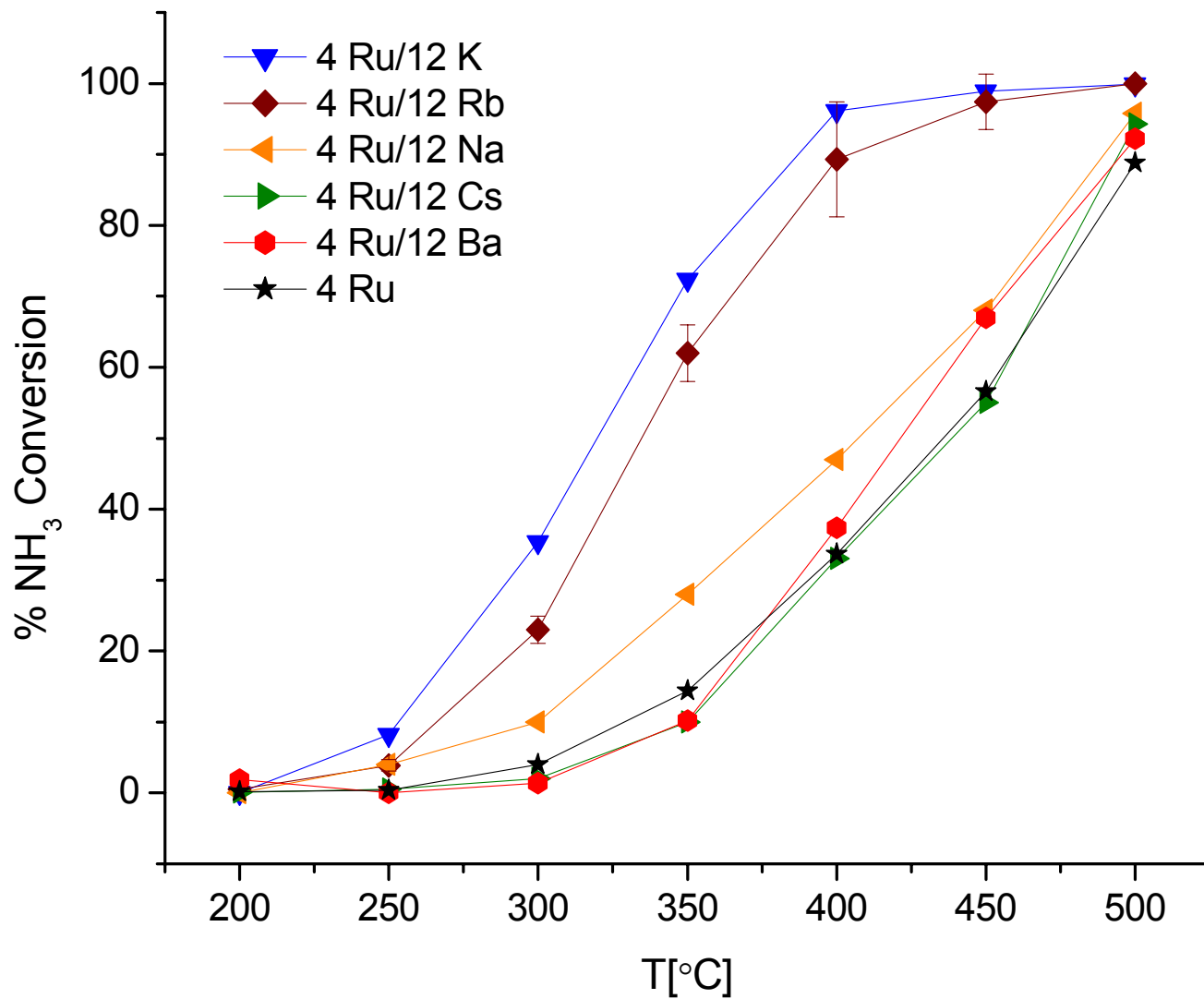


Term	Coefficients
Constant	35.7
Ba	-1.8
K	22.3
Cs	-2.6
Ba*Ba	-0.5
K*K	5.6
Cs*Cs	-3.0
Ba*K	-0.9
Ba*Cs	-1.0
K*Cs	-4.7

$$R = C + \alpha_1(Ba) + \alpha_2(K) + \alpha_3(Cs) + \dots + \beta_1(Ba)^2 + \beta_2(K)^2 + \dots + \lambda_1(Ba * K) + \lambda_2(Ba * Cs) + \dots$$

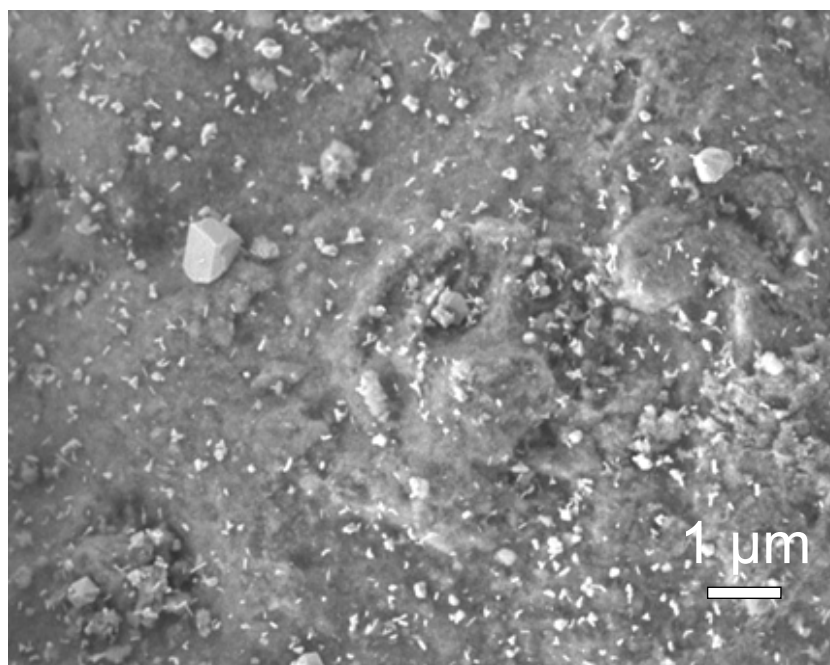


# Promotion of Ru Catalyst

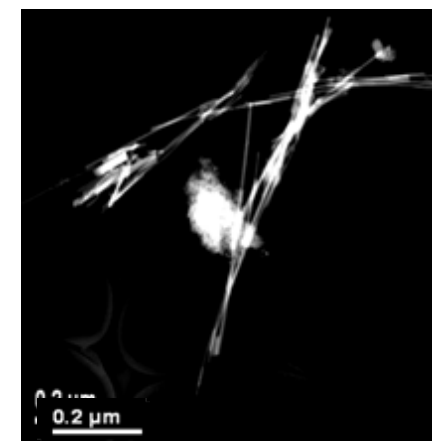
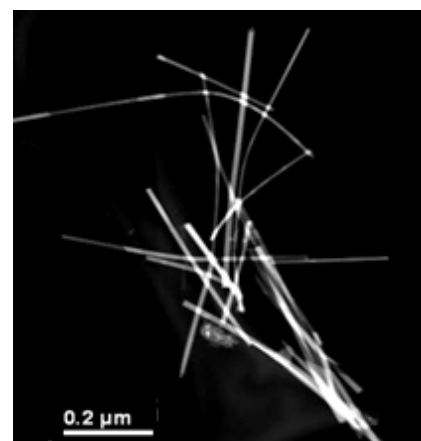
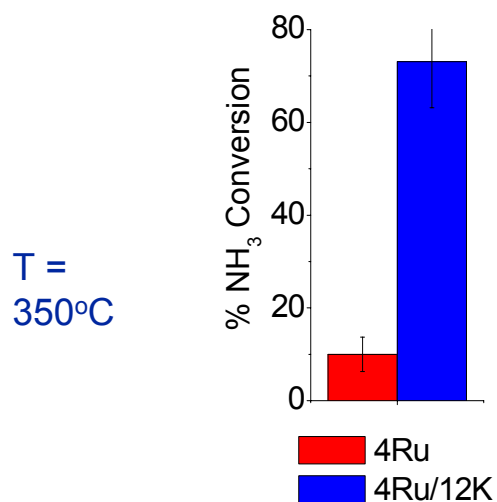
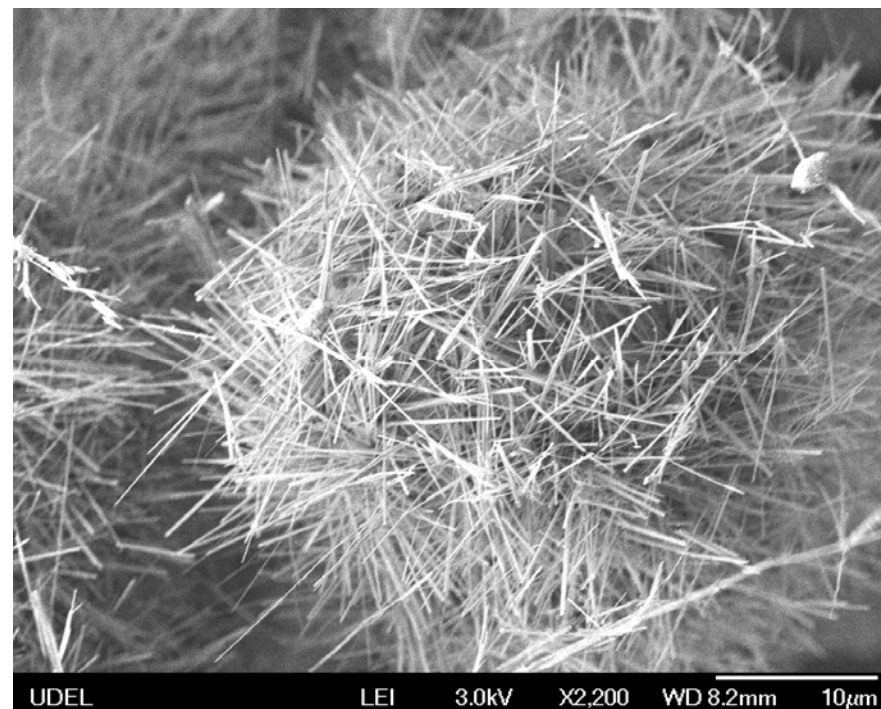


# K-Promotion: Effect on Morphology

4Ru /  $\gamma$ -Al<sub>2</sub>O<sub>3</sub>



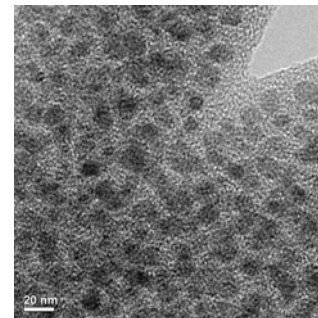
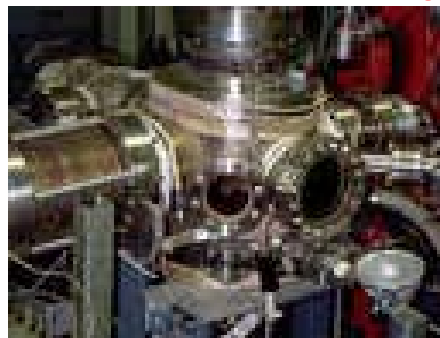
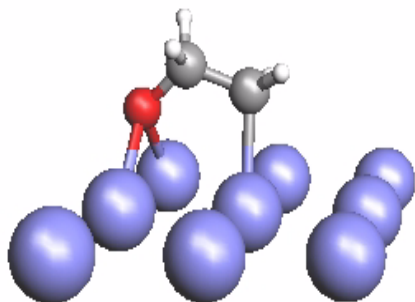
4Ru-12K /  $\gamma$ -Al<sub>2</sub>O<sub>3</sub>



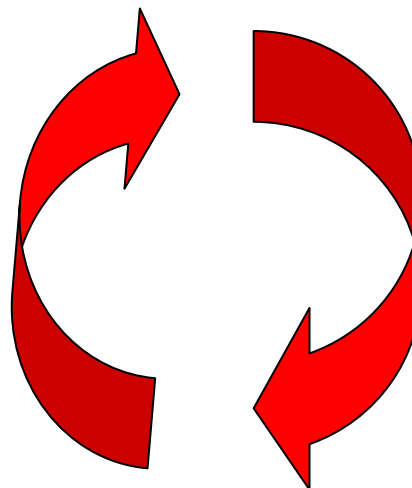
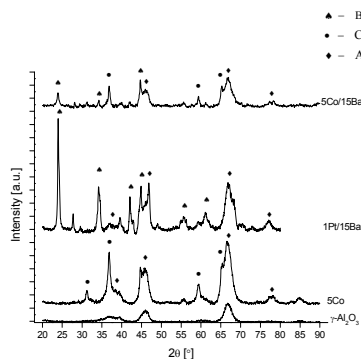
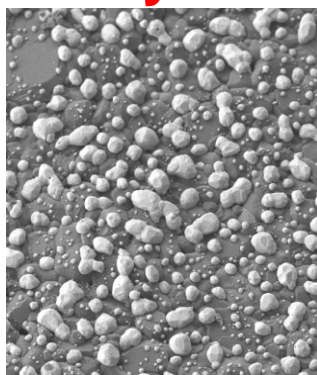
# Our Research Philosophy - HT Catalytic Science

R. J. Hendershot,  
C.M. Snively,  
and J. Lauterbach,  
Chemistry –  
A European Journal,  
11; 806-814, 2005

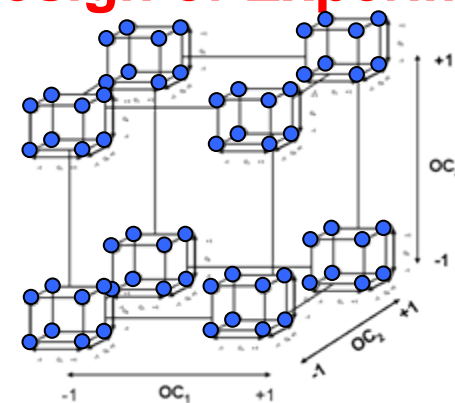
## Modeling + UHV + Spectroscopy + Synthesis



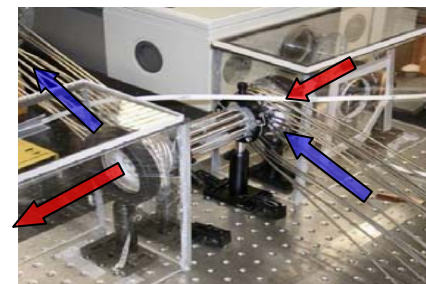
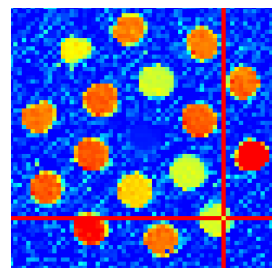
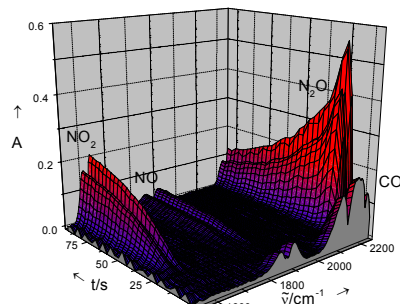
## Catalyst Characterization



## Design of Experiment



## Reactor

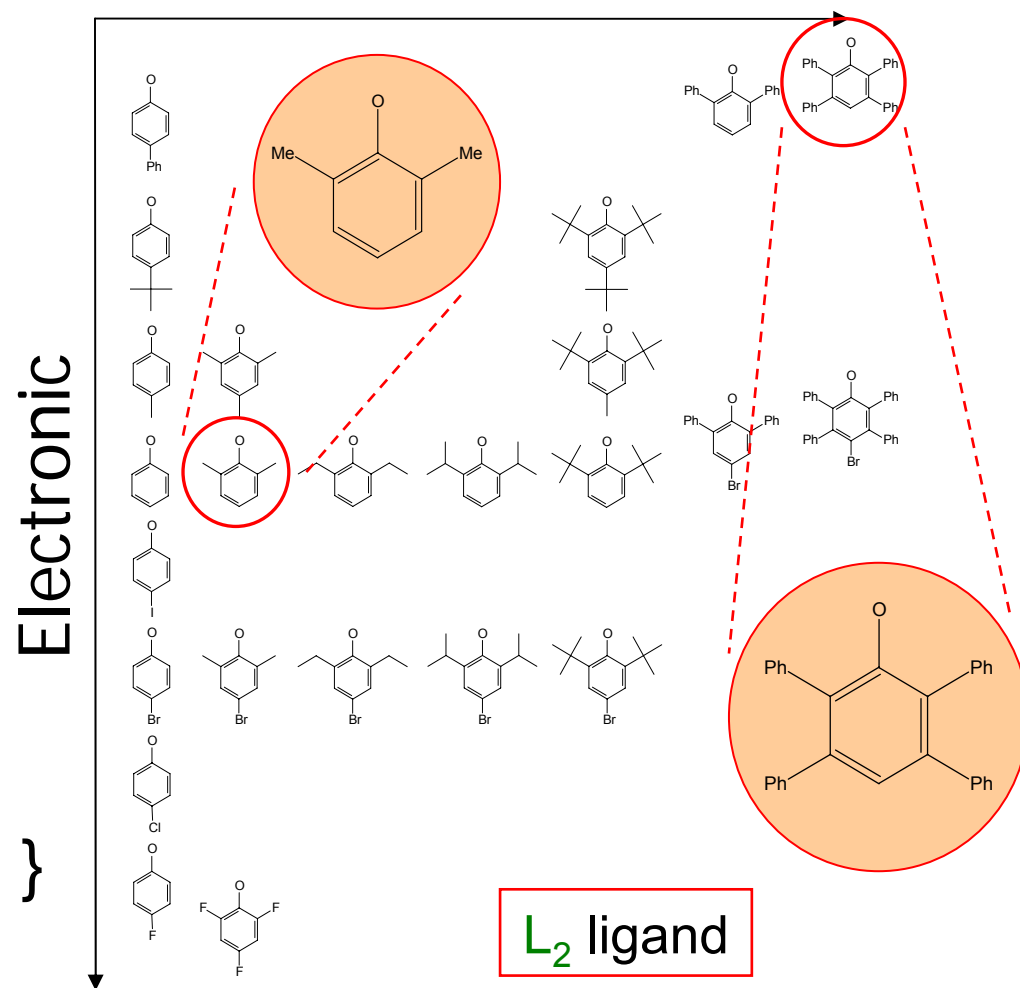
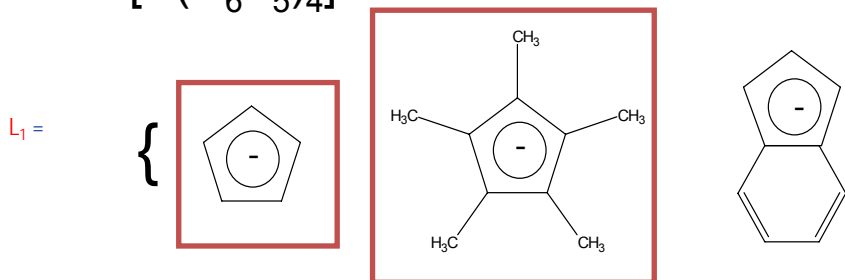
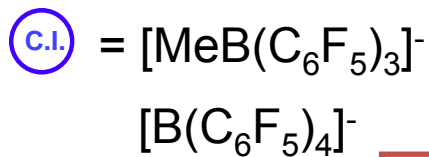
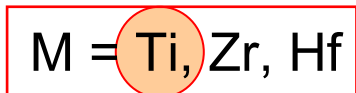
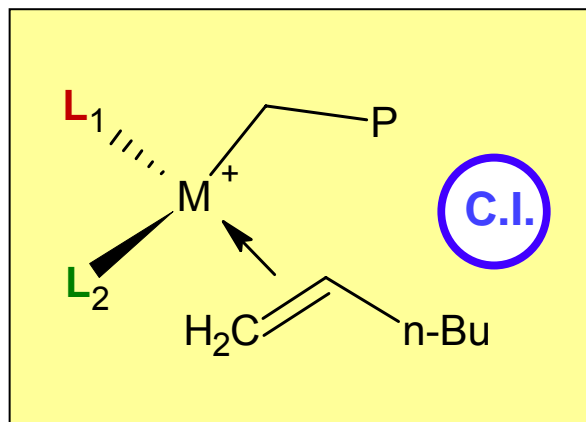


# 1- Hexene Polymerization by Titanium Catalysts

## with Phenoxy based ligands

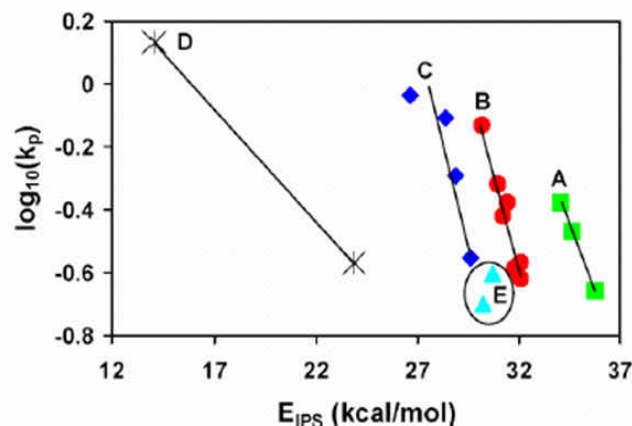
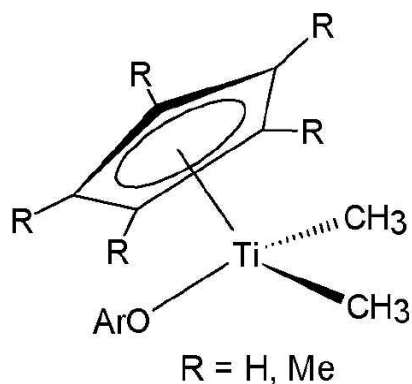
A large number of available substituted phenols allow tunability of steric and electronic variation of the catalyst.

Steric



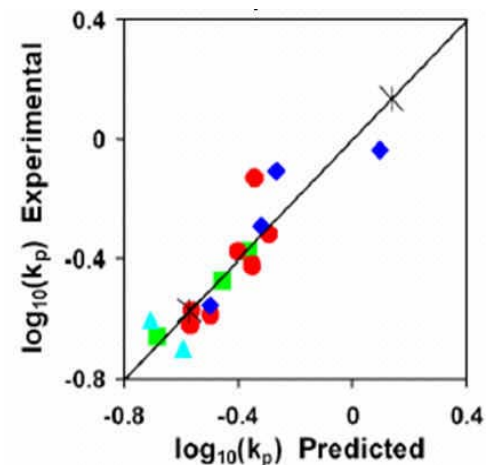


# Structure-Activity Correlation in Single-Site, Aryl-Oxide, Olefin Polymerization Catalysts



Family-dependent correlation of propagation rate versus  $E_{IPS}$

Universal correlation

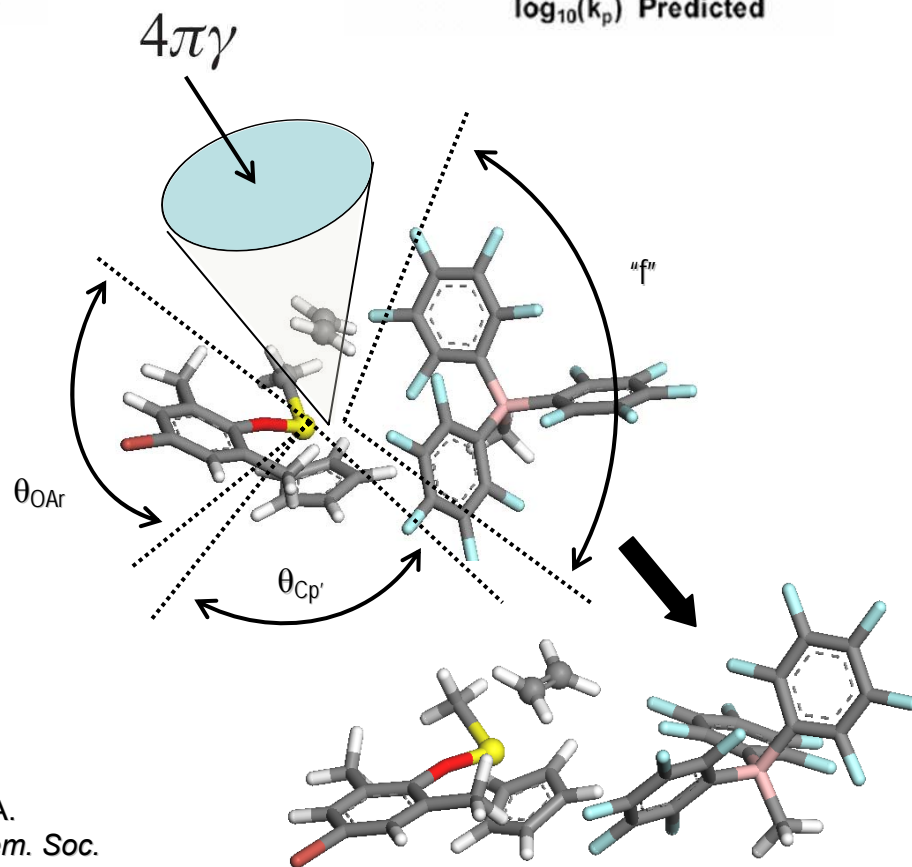


DFT-Based Structure Model: The solid-angle available for monomer approach ( $4\pi\gamma$ ) to Ti-center is related to the OAr- and Cp'-ligand exclusion cone-angles ( $\theta$ ) and a steric factor ( $f$ ) related to the counter-anion exclusion.

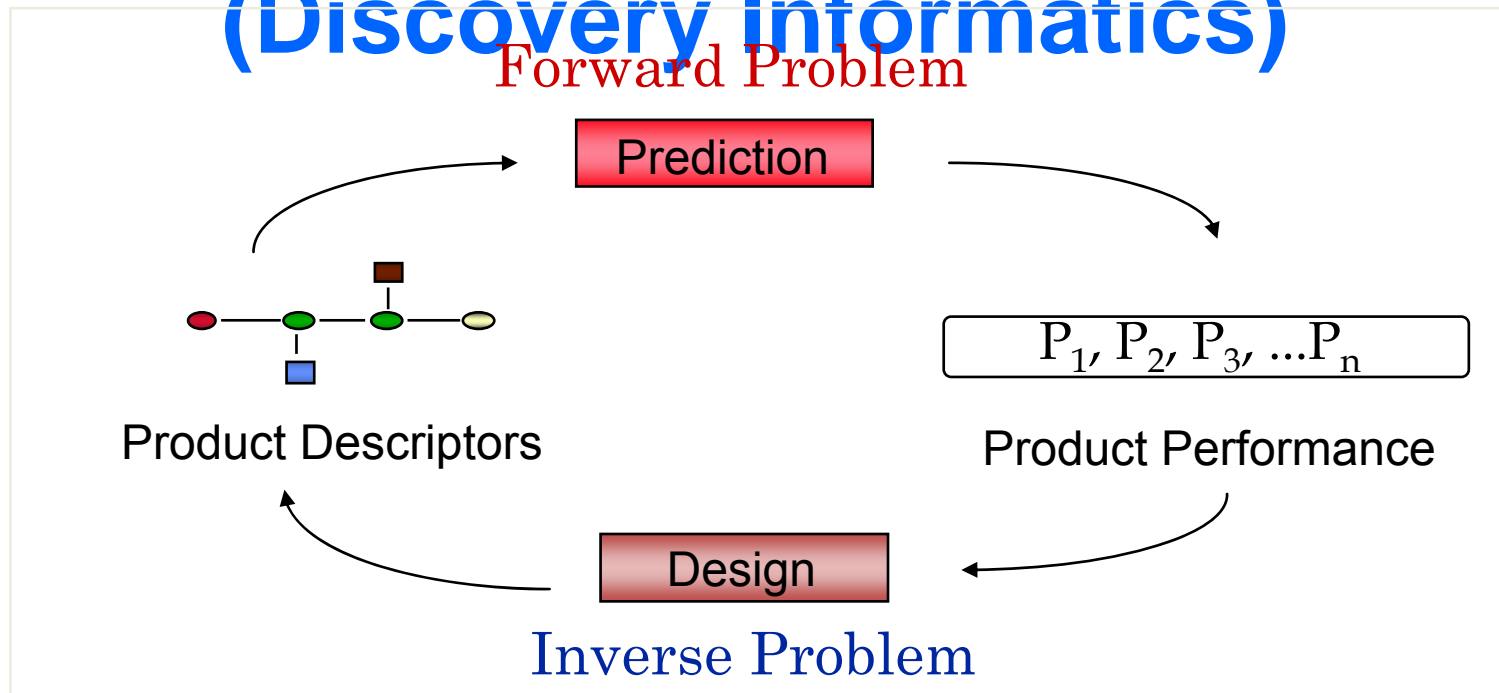
$$\gamma = 1 - \sin^2(\theta_{Cp'}/4) - \sin^2(\theta_{OAr}/4) - f$$

The propagation rate constant correlates with ion-pair separation energy ( $E_{IPS}$ ) in a universal manner:

$$k_{pred} = k_0 e^{-E_a/RT} = \gamma a_0 e^{-E_0/RT} e^{-\alpha E_{IPS}/RT}$$



# Computer-aided Product Formulation/Design (Discovery Informatics)

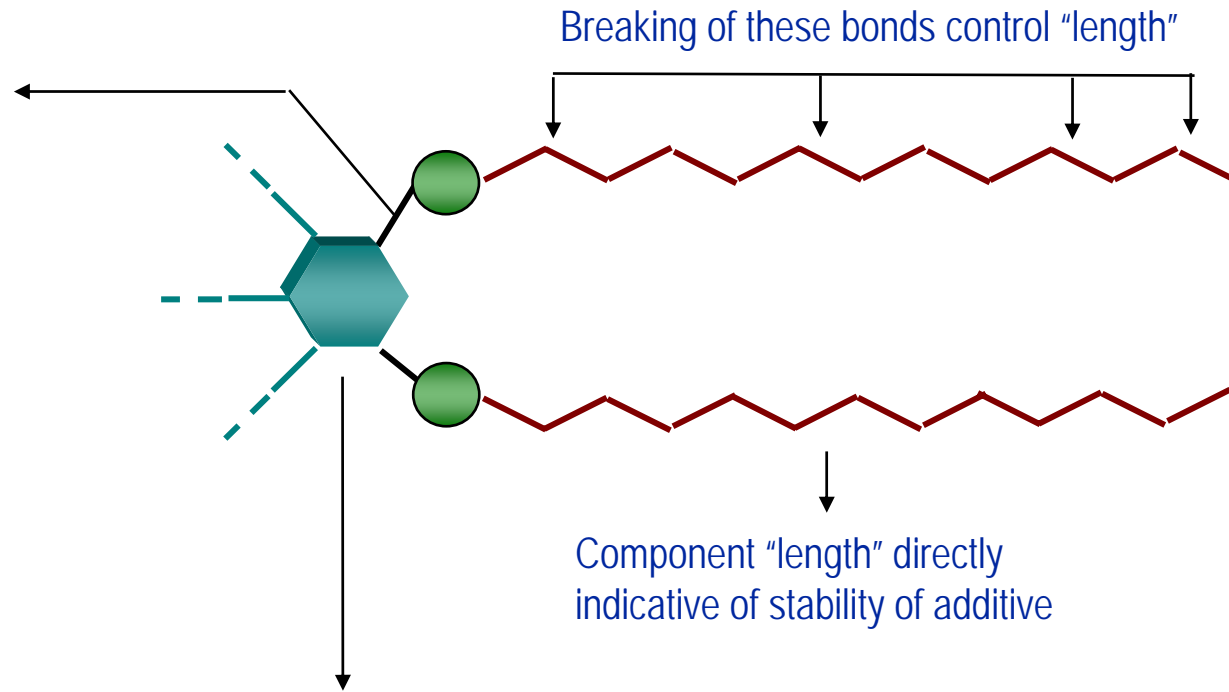


- **Forward Problem: Prediction**
  - Estimate Product Performance from Descriptors
    - Quantitative forward model
- **Inverse Problem: Design**
  - Determine a set of products that satisfy desired performance criteria
    - Guided stochastic search (e.g. genetic algorithm)

# *Search for New Surfactant Molecules*

## *Forward model: First-Principles Model of Additive Degradation + neural net model of engine performance*

Breakage of this bond removes "dirt" carrying capacity totally



Chemical nature of this component (polar/non-polar) controls "dirt" removing capacity

The first-principle model tracks the structural distribution of fuel-additive with time due to reactive degradation

# Results of GA Inverse Search

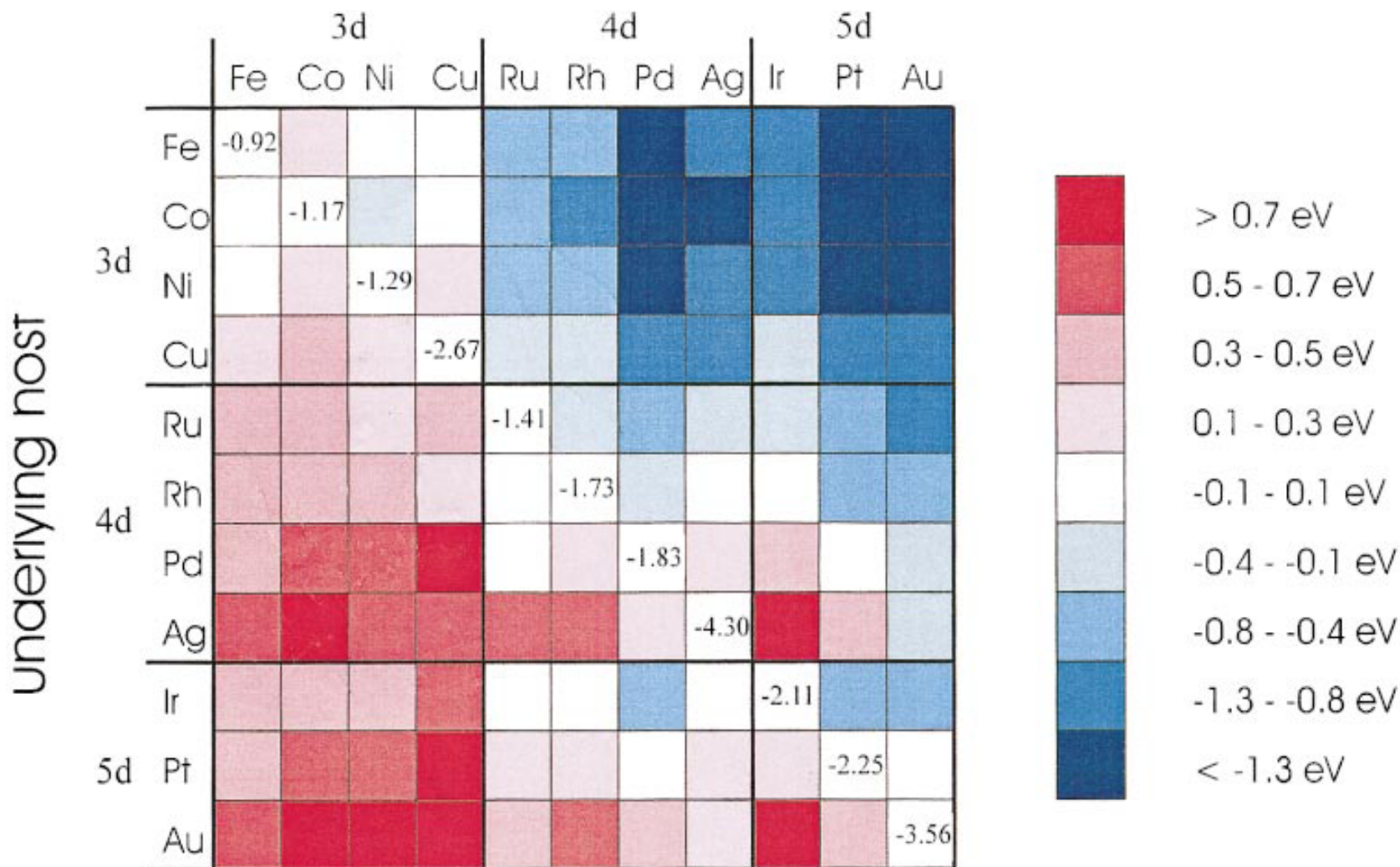
**Objective:**  
*Determine a structure with IVD < 10 mg*

**Population Size: 25;**  
**Generations: 25**

Run	Rank/Identifier	Fitness	Predicted IVD (PLS-NN Model)	Structural Description
I	1, I-1	0.997	11.4 mg	Novel Structure. Infrequently used linker.
	2, I-2	0.996	11.5 mg	Novel Structure. Same tails as best structure, different heads and linkers
	6, I-6	0.993	12.0 mg	Variation of structure found in the BMW database. Same head and linkers, different tails
II	1, II-1	0.999	10.1 mg	Novel Structure. Different from I-1 . Infrequently used linker component.
	2, II-2	0.989	12.6 mg	Slight variation of additive structure found in BMW and HONDA databases. Different tails but same head and linker
	4, II-4	0.983	13.2 mg	Minor variation of structure II-2 above. Slight modification of the head
III	1, III-1	1.00	8.9 mg	<b>Novel Structure</b> . Different from I-1 and II-1 . Commonly used components
	2, III-2	0.994	11.9 mg	Variation of III-1 . One linker and tail modified.
	3, III-3	0.993	12.1 mg	Variation of structure II-2 above. Slight modification of head. A linker and tail inserted.

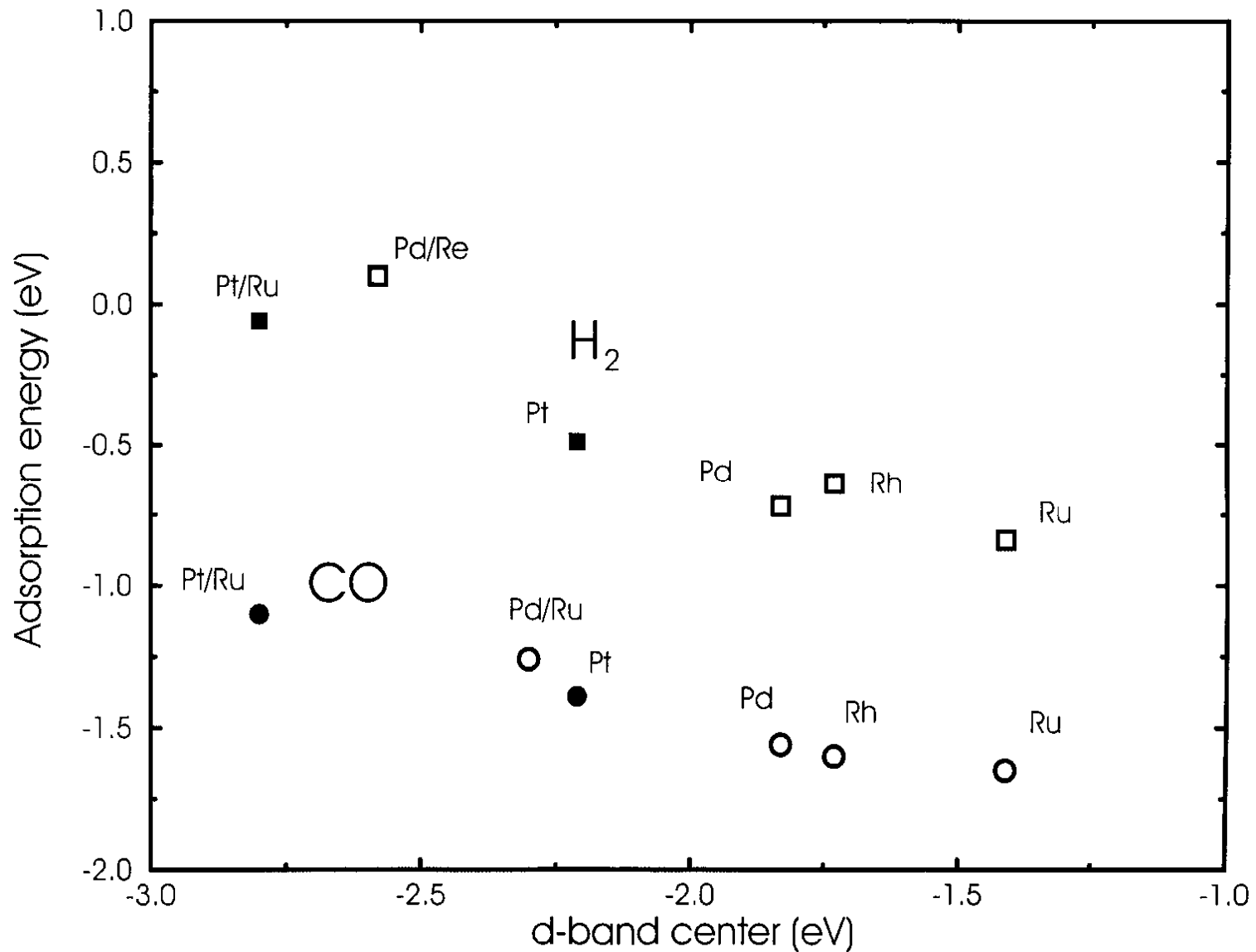


# Norskov d-band Center Descriptor (DFT)



E. Christoffersen, P. Liu, A. Ruban, H.L. Skiver, and J. K. Norskov, "Amode Materials for Low-Temperature Fuel Cells: A Density Functional Theory Study". *J. Catal.*, **199**, 123-131 (2001).

# Correlation of $\Delta H_{\text{ads}}$ with d-Band Center



E. Christoffersen, P. Liu, A. Ruban, H.L. Skiver, and J. K. Norskov, "Amode Materials for Low-Temperature Fuel Cells: A Density Functional Theory Study". *J. Catal.*, **199**, 123-131 (2001).

# **Drug Discovery Approach to Breakthroughs in Batteries**

**Informatics in Drug Discovery** Ernst R. Dow, Eli Lilly

**Electrochemical energy storage and extended-range electric vehicles** Mark Verbrugge, GM

**Data handling and informatics tools for model-based discovery**  
James Caruthers, Purdue

**New High Energy/Power Devices** Ralph J. Brodd, Broddarp

**High-throughput ab-initio computing and data mining methods for the prediction of crystal structure** Gerbrand Ceder, MIT

**Materials Informatics: An “omics” approach to materials based design for battery technology** Krishna Rajan, Iowa State

# **Discussion at dinner tonight**

## **Breakout Groups (Tuesday)**

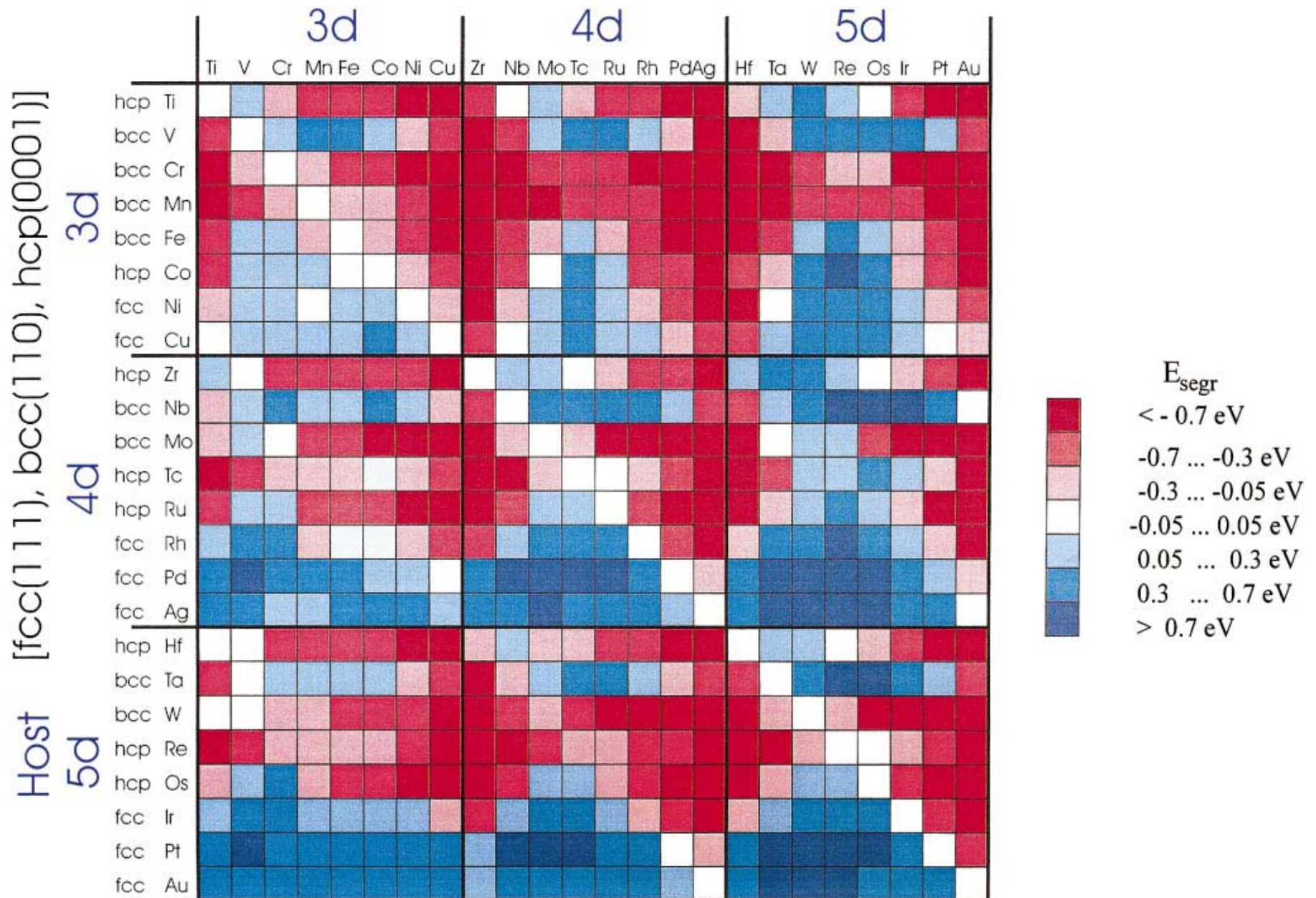
- **Two discussion sessions**
- **One writing session**

## **Objective**

- **Report:**
  - Define and justify opportunities for significant advancement in battery technology.**
  - Suggest guidelines for program structure**
- **Website: Maintain as a resource for the community. Start by uploading slides of talks**



# Computed surface segregation energies



E. Christoffersen, P. Liu, A. Ruban, H.L. Skiver, and J. K. Norskov, "Amode Materials for Low-Temperature Fuel Cells: A Density Functional Theory Study". *J. Catal.*, **199**, 123-131 (2001).