## **Incompatible Chemicals List**

A wide variety of chemicals react dangerously when mixed with certain other materials. Some of the more widely used incompatible chemicals are given below, but the absence of a chemical from this list should not be taken to indicate that it is safe to mix it with any other chemical.

- acetic acid: chromic acid, ethylene glycol, nitric acid, hydroxyl compounds, perchloric acid, peroxides, permanganates
- acetone: concentrated sulphuric and nitric acid mixtures
- acetylene: chlorine, bromine, copper, fluorine, silver, mercury
- alkali and alkaline earth metals: water, chlorinated hydrocarbons, carbon dioxide, halogens, alcohols, aldehydes, ketones, acids
- **aluminium** (**powdered**): chlorinated hydrocarbons, halogens, carbon dioxide, organic acids.
- anhydrous ammonia: mercury, chlorine, calcium hypochlorite, iodine, bromine, hydrofluoric acid
- **ammonium nitrate**: acids, metal powders, flammable liquids, chlorates, nitrites, sulphur, finely divided organic combustible materials
- aniline: nitric acid, hydrogen peroxide
- arsenic compounds: reducing agents
- azides: acids
- **bromine**: ammonia, acetylene, butadiene, hydrocarbons, hydrogen, sodium, finely-divided metals, turpentine, other hydrocarbons
- calcium carbide: water, alcohol
- calcium oxide: water
- carbon, activated: calcium hypochlorite, oxidizing agents
- **chlorates**: ammonium salts, acids, metal powders, sulphur, finely divided organic or combustible materials
- **chromic acid**: acetic acid, naphthalene, camphor, glycerin, turpentine, alcohols, flammable liquids in general
- **chlorine**: see bromine
- chlorine dioxide: ammonia, methane, phosphine, hydrogen sulphide
- copper: acetylene, hydrogen peroxide
- cumene hydroperoxide: acids, organic or inorganic
- cyanides: acids
- **flammable liquids**: ammonium nitrate, chromic acid, hydrogen peroxide, nitric acid, sodium peroxide, halogens
- hydrocarbons: fluorine, chlorine, bromine, chromic acid, sodium peroxide
- hydrocyanic acid: nitric acid, alkali
- hydrofluoric acid: aqueous or anhydrous ammonia
- **hydrogen peroxide**: copper, chromium, iron, most metals or their salts, alcohols, acetone, organic materials, aniline, nitromethane, flammable liquids, oxidizing gases
- hydrogen sulphide: fuming nitric acid, oxidizing gases

- hypochlorites: acids, activated carbon
- iodine: acetylene, ammonia (aqueous or anhydrous), hydrogen
- mercury: acetylene, fulminic acid, ammonia
- mercuric oxide: sulphur
- nitrates: sulphuric acid
- **nitric acid (conc.)**: acetic acid, aniline, chromic acid, hydrocyanic acid, hydrogen sulphide, flammable liquids, flammable gases
- oxalic acid: silver, mercury
- perchloric acid: acetic anhydride, bismuth and its alloys, ethanol, paper, wood
- peroxides (organic): acids, avoid friction or shock
- phosphorus (white): air, alkalies, reducing agents, oxygen
- potassium: carbon tetrachloride, carbon dioxide, water
- potassium chlorate: acids
- potassium perchlorate: acids
- **potassium permanganate**: glycerin, ethylene glycol, benzaldehyde, sulphuric acid
- selenides: reducing agents
- silver: acetylene, oxalic acid, tartaric acid, ammonium compounds, fulminic acid
- sodium: carbon tetrachloride, carbon dioxide, water
- **sodium nitrate**: ammonium salts
- **sodium peroxide**: ethanol, methanol, glacial acetic acid, acetic anhydride, benzaldehyde, carbon disulphide, glycerin, ethylene glycol, ethyl acetate, methyl acetate, furfural
- sulphides: acids
- **sulphuric acid**: potassium chlorate, potassium perchlorate, potassium permanganate (or compounds with similar light metals, such as sodium, lithium, etc.)
- tellurides: reducing agents
- **zinc powder**: sulphur