Mollie Burgoon 5.12 exam #3

### Alkene synthesis

<u>Dehydrohalogenation</u> strong base

#### Dehalogenation of a vicinal dibromide

NaI, acetone

#### **Reactions of alkenes**

Electrophilic addition of HX mark alkyl halide HX and polar solvent

Free radical addition of HBr antimark alkyl halide HBr, ROOH, heat

 $\frac{Electrophilic\ H_20\ addition}{mark}$  mark alcohol  $H^+,\ H_20$ 

Oxymercuration/demercuration mark alcohol

Hg(OAc)<sub>2</sub>/H<sub>2</sub>0, NaBH<sub>4</sub>

Hydroboration/oxidation antimark alcohol BH<sub>3</sub>/THF, H<sub>2</sub>0<sub>2</sub>/KOH

Hydrogenation alkane H<sub>2</sub> and Pt/C

<u>Halogenation</u> vicinal dihalide X<sub>2</sub>

Hydrohalogenation mark (OH) halohydrin (alcohol moiety) X<sub>2</sub>, H<sub>2</sub>0

Syn dihydroxylation dialcohol OsO<sub>4</sub> or KmnO<sub>4</sub>, H<sub>2</sub>O<sub>2</sub>

Epoxidation/hydrolysis anti dialcohol MCPBA, H<sub>2</sub>/H<sub>2</sub>O

Ozonolysis cleaved double bond

O<sub>3</sub>, (CH<sub>3</sub>)<sub>2</sub>S

Cationic polymerization BF<sub>3</sub> or ROOR

If you want	and you have	reagents	+/-	
alcohol	alkene	$H^{+}, H_{2}0$	+	hydration
	alkene	Hg(OAc) <sub>2</sub> /H <sub>2</sub> 0, NaBH <sub>4</sub>	+	oxymercuration
	alkene	BH <sub>3</sub> /THF, H <sub>2</sub> 0 <sub>2</sub> /KOH	-	hydroboration
	alkene	OsO <sub>4</sub> , H <sub>2</sub> O <sub>2</sub>		syn dihydroxylation
	alkene	MCPBA, H <sub>2</sub> /H <sub>2</sub> O		epoxidation/hydrolysis
alkyl halide	alkene	HX, polar solvent	+	electrophilic HX addition
	alkene	HBR, ROOH, $\Delta$	-	free radical addition
vicinal dihalide	alkene	$X_2$		halogenation
halohydrin	alkene	X <sub>2</sub> , H <sub>2</sub> 0	+	halohydrogenation
alkane	alkene	H <sub>2</sub> and Pt/C		hydrogenation
alkene	alkyl halide	strong base		dehydrohalogenation
	vicinal dibromide	NaI, acetone		dehalogenation
	alcohol	strong acid (H <sub>2</sub> SO <sub>4</sub> )		dehydration of alcohol

#### **Carbene reactions**

acts like a radical prone to forming 3-membered rings  $I_2CH_2 + Zn \text{ or } Cu$ 

**Expoxidation of alkenes** 3-membered ring uses MCPBA (metachloroperbenzoic acid)

## Reactions of alkynes

Lindlar's catalyst Pd + BaSO4 + quinine turn alkyne into alkene

Acetylide ion deprotonate -≡-H to get -≡: only way to make a C-C bond NaNH<sub>2</sub>

# Look at synthesis of alkynes!!