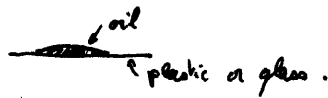
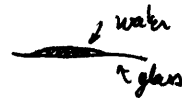
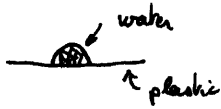


② Contact with a solid

experiments:



2.1 Wetting parameter



$$S = \sigma_{SV} - (\sigma_{LV} + \sigma_{SL})$$

$S > 0 \Rightarrow$ total wetting (film of liquid)

$S < 0 \Rightarrow$ partial wetting (droplet).

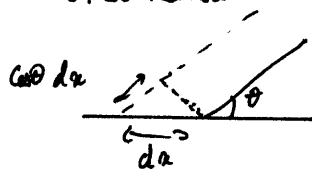
2.2 Partial wetting



θ ($\propto R$) to minimize E_s with constant V !

\rightarrow geometry argument works but long.

other method:



slight perturbation close to equilibrium

$$dE_s = (\sigma_{SL} - \sigma_{SV}) dx + \sigma_{LV} \cos \theta dx$$

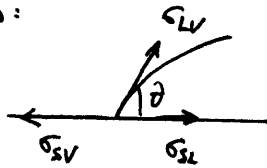
$$dE_s = 0 \text{ (equilibrium)}$$

$$\cos \theta = \frac{\sigma_{SV} - \sigma_{SL}}{\sigma_{LV}}$$

Young relation (1805)

\propto relation for macroscopic \rightarrow only for perfect surface.

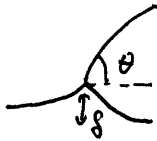
other derivation: forces:



→ projection on the horizontal ⇒ Young relation.

vertical? → pinching of the surface.

→ slide.



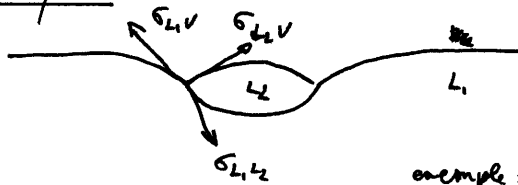
$$\sigma_{LV} \lambda \sin \theta \sim E \delta$$

↓
Young modulus (elastic)

$$\delta \sim \frac{\sigma}{E} \quad E \sim \text{GPa} \Rightarrow \delta \sim 1 \text{ nm.}$$

↳ only visible for soft solids.

case liquid/liquid:



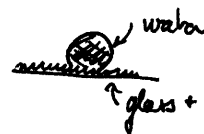
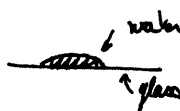
example: oil on water.

$$\text{equilibrium: } \vec{\sigma}_{L1V} + \vec{\sigma}_{L2V} + \vec{\sigma}_{L1L2} = \vec{0}.$$

~~surface~~

surface treatment

experiment:



$$\cos \theta = \frac{\sigma_{SV} - \sigma_{SL}}{\sigma_{LV}} \rightarrow \text{treatment} \Rightarrow \sigma_{SV} \downarrow, \sigma_{SL} \uparrow$$

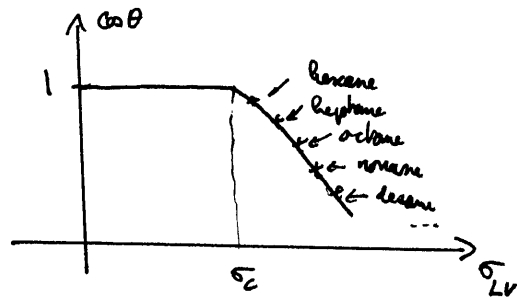
$$\Rightarrow \cos \theta \downarrow, \theta \uparrow$$

↳ pb: measures of σ_{SV} , σ_{SL} ?

2.3 Zisman criterion

surface tension of a solid! only $\sigma_{SV} - \sigma_{SL}$ (from σ_{LV} and θ)

Zisman method: $\cos \theta$ versus σ_{LV} (chemicals of the same family).



if $\sigma < \sigma_c \Rightarrow$ total wetting.

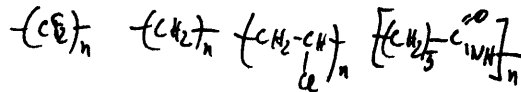
semi-empirical: fine for non polar liquids

slide.

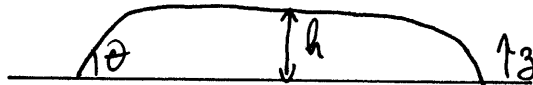
Typical values for σ_c :

material	Teflon	PE	PVC	Nylon
σ_c (mN/m)	~ 18	~ 30	~ 40	~ 50

more polar $\Rightarrow \sigma_c \uparrow$

2.4 From droplets to puddles

experiment: syringe.



thickness h ? \rightarrow gravity / surface tension

gravity: $dE_p = \rho g A dy \Rightarrow E_p = \frac{1}{2} \rho g h^2 A$

surface energy: $E_s = (\sigma_{LV} + \sigma_{SL} - \sigma_{SV}) A = -S A$. ($S < 0$, partial wetting)

$$E = -S A + \frac{1}{2} \rho g A h^2$$

$$V = A h \Rightarrow \text{constant.}$$

$$E/V = -\frac{S}{h} + \frac{1}{2} \rho g h$$

$$E/V \text{ min} \rightarrow \frac{dE/V}{dh} = 0 \rightarrow \frac{\sigma}{h^2} + \frac{1}{2} \rho g = 0$$

$$h = \sqrt{\frac{-2\sigma}{\rho g}} = \sqrt{\frac{2 \sigma_{LV} (-\cos \theta)}{\rho g}}$$

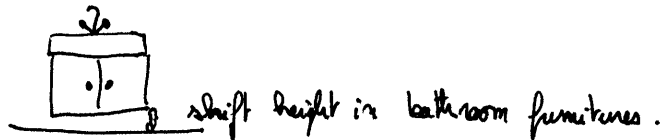
$$h = 2 \cos \frac{\theta}{2} \cdot \sqrt{\frac{\sigma}{\rho g}}$$

↓
capillary length l_c (or κ^{-1}).

typical values: oil: $l_c \sim 1 \text{ mm}$.

water $l_c = 2.7 \text{ mm}$.

$h \nearrow$ if $\theta \nearrow \rightarrow$ lower for $\theta = \pi \rightarrow$ water $\sim 5 \text{ mm}$.



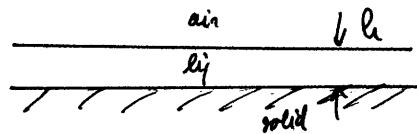
2.3 Total wetting

$\sigma > 0 \rightarrow$ liquids with low σ (Zisman).

↳ how far can it spread?



Van der Waals interactions for a thin film:



{ so close that the solid "feels" the upper interface.

typical cut-off: 100 nm

$$\rightarrow E/h = \sigma_{SL} + \sigma_{SV} + \Sigma(h)$$

$$E(h) = \frac{H}{12.7 h^2} \rightarrow \text{Hamaker constant } \sim \text{contrast in the materials polarizabilities.}$$

$H \sim 10^{-49} \cdot 10^{-20} \text{ J}$ usually > 0 (solid more polarizable)
 < 0 (-veff).

Final thickness of the pancake:

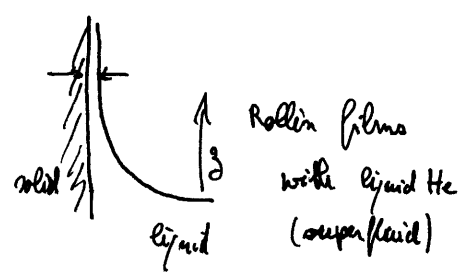
$$E/V = -\frac{S}{a} + \frac{H}{4\pi a^3}$$

$$E/V)_{\min} \rightarrow \frac{S}{a^2} = \frac{H}{4\pi a^4} \Rightarrow \boxed{a = \sqrt{\frac{H}{4\pi S}}}$$

→ slide: measure AN Cezalbet.

Disjoining pressure

$$P_d = \frac{dZ}{da} = -\frac{H}{6\pi a^3}$$

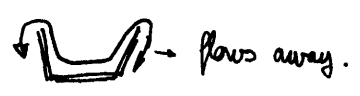


hydrostatic ↓ van der Waals.

$$\rightarrow -\rho g = -\frac{H}{6\pi a^3}$$

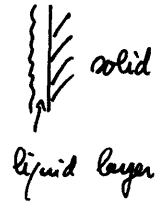
$$\boxed{a = \left(\frac{H}{6\pi\rho g}\right)^{1/3}}$$

→ Pb: filling a container with liquid He:



→ partial wetting on G.

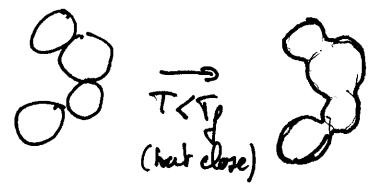
Surface melting:



$T < T_f \Rightarrow$ should not melt, but $E \Rightarrow$ if the solid is wetted by its own liquid \Rightarrow surface melting.

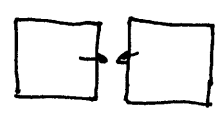
\hookrightarrow thin film (molecular thickness) gets thick when $T \rightarrow T_f$.

Thin part responsible for sintering:



\hookrightarrow question from Faraday: how to make snow balls below 0°C !

\hookrightarrow "adhesion" of ice cubes:



\hookrightarrow layer of liquid solidifies (no more interface).