

Modeling radon daughter deposition rates for low background detectors

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The Problem

- Many next-generation detectors such as dark matter detectors and neutrinoless double-beta decay detectors require record low levels of background radiation
- Radon (most commonly ^{222}Rn) is a radioactive daughter of uranium found in the earth and is present in air
 - Daughters of ^{222}Rn are deposited on materials exposed to air
 - Long-lived daughters, particularly ^{210}Pb (with a half-life of 22 years) may present a long term source of background radiation above the acceptable threshold (for example, the threshold for miniCLEAN is $1\alpha/\text{m}^2/\text{day}$)

The Goal

- A better understanding of radon daughter deposition may allow for procedures to be developed that will minimize the amount of background radiation from this source
- The goal is to develop a model of radon daughter deposition so it can be understood as a function of several different environmental variables, including radon concentration, humidity, particle count, temperature, and so on

Deposition Setup

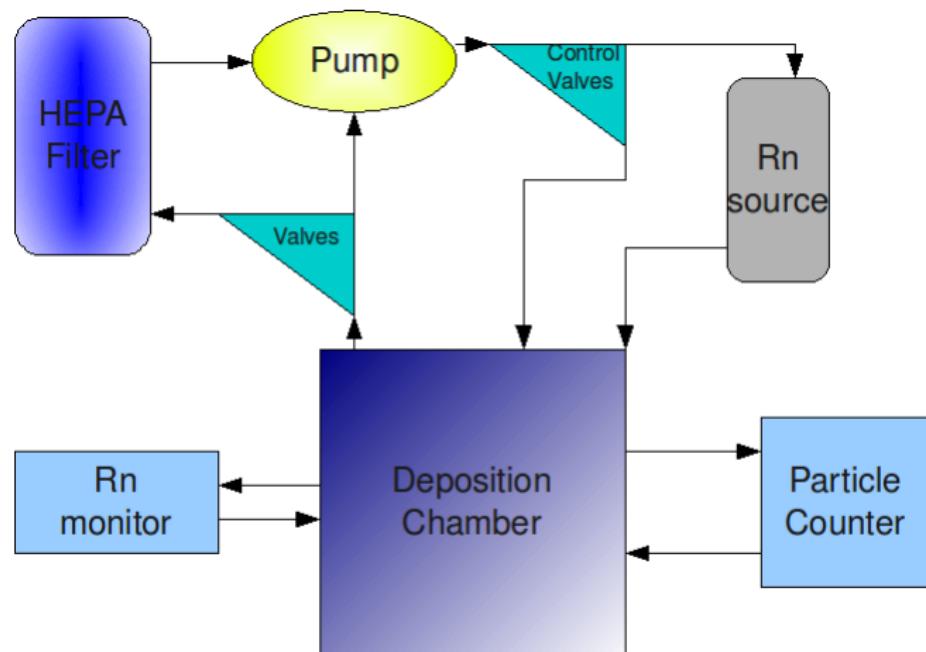


Figure: Diagram of deposition setup

Deposition Setup cont.



Figure: Deposition setup

Detector Setup

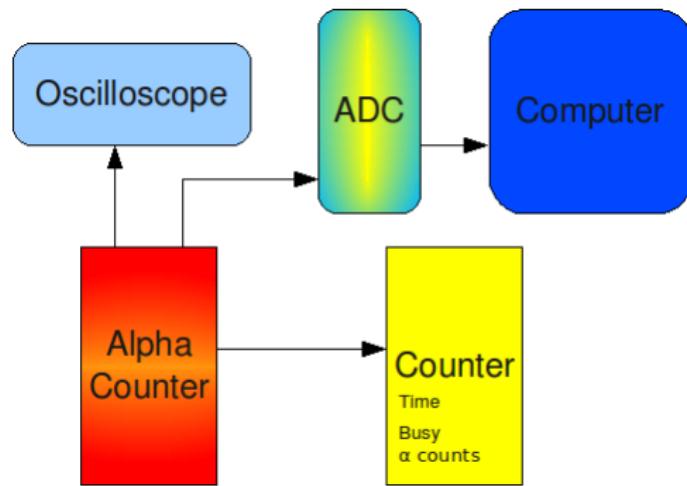


Figure: Diagram of detector and data acquisition

α Detector Geometric Efficiency

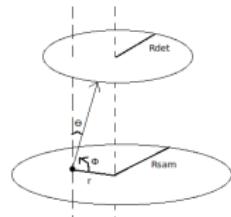


Figure: Model of α particles leaving sample and hitting detector

- What percentage of the α particles that leave the sample will hit the detector?
- Conditions for the α particle to hit the detector:
 - distance traveled by the particle parallel to the plates is

Distance Traveled

$$a = h \tan \theta \quad (1)$$

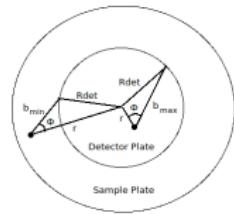


Figure: Top-down view

Geometric Efficiency cont.

- distance to edge of detector plate is

Maximum Distance

$$b_{max} = r \cos \phi + \sqrt{R_{det}^2 - r^2 \sin^2 \phi} \quad (2)$$

- distance to reach the detector plate

Minimum Distance

$$b_{min} = r \cos \phi - \sqrt{R_{det}^2 - r^2 \sin^2 \phi} \quad (3)$$

- So in order for the particle to hit the detector, a must be less than b_{max} , non-negative, and greater than b_{min}
- Running a Monte Carlo simulation with randomized values for ϕ, θ , and r yields a geometric efficiency of 0.246944

α Detector Sampling Frequency

- What is the maximum rate at which the detector can detect α particles?
- Is the detector likely to miss a significant number of α particles if the radon daughters decay too quickly?

α Detector Sampling Frequency

- What is the maximum rate at which the detector can detect α particles?
- Is the detector likely to miss a significant number of α particles if the radon daughters decay too quickly?
- Found that the detector does not start missing pulses until they are coming in at a frequency of approximately 40 kHz

α Detector Sampling Frequency cont.

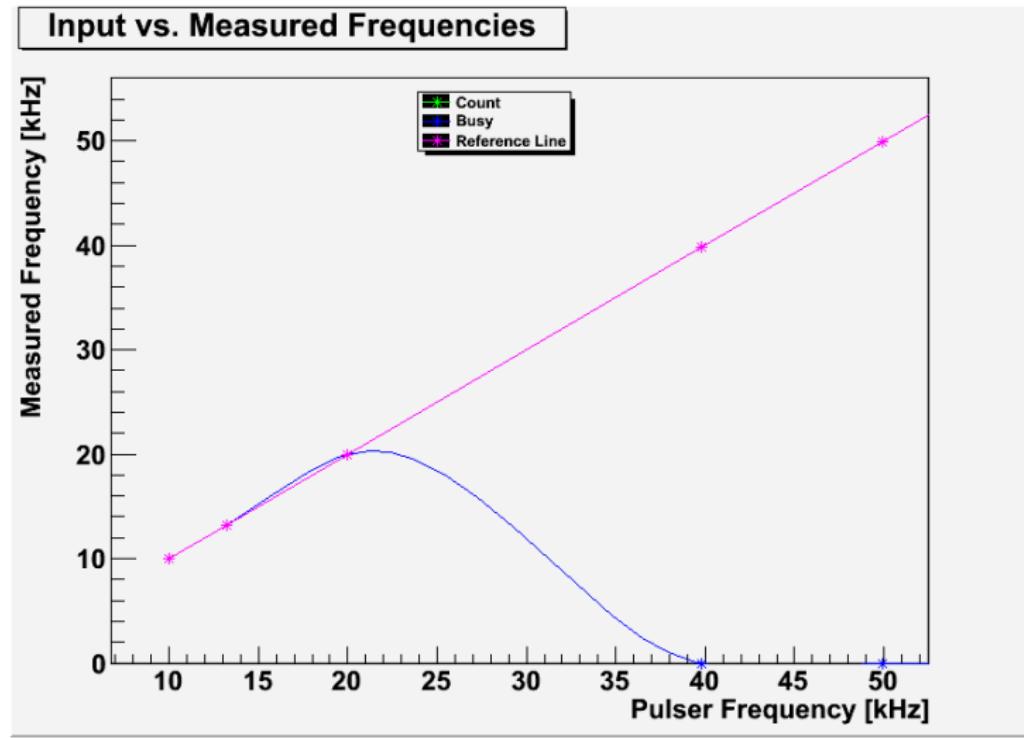


Figure: Pulse Generator vs. Detected Frequencies

ADC Sampling Frequency

- What is the maximum rate at which the ADC can collect data from the α detector?
- Is data likely to be lost due to restrictions on the ADC?

ADC Sampling Frequency

- What is the maximum rate at which the ADC can collect data from the α detector?
- Is data likely to be lost due to restrictions on the ADC?
- Used a random generator to send pulses to ADC and compared the measured frequencies to the input frequencies

ADC Sampling Frequency cont.

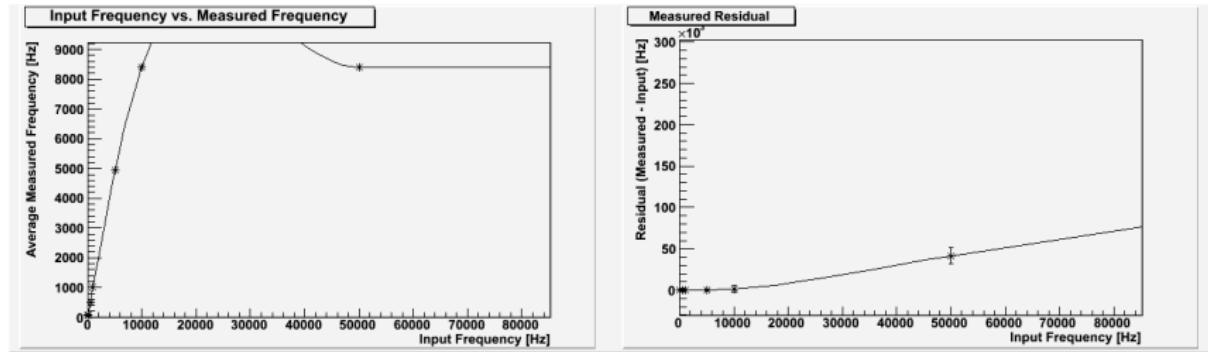
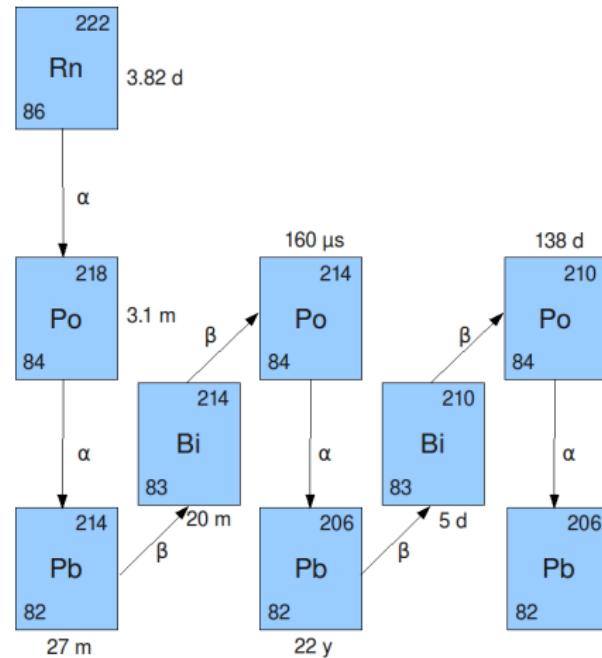


Figure: Pulse Generator vs. Measured Frequencies

Decay Chain



The Model

$N_1 = {}^{218}\text{Po}$, $N_2 = {}^{214}\text{Pb}$, $N_3 = {}^{214}\text{Bi}$, $N_4 = {}^{214}\text{Po}$

Deposition

$$\frac{dN_1}{dt} = C_1 D_1 - \lambda_1 N_1 \quad (4)$$

$$\frac{dN_2}{dt} = C_2 D_2 - \lambda_2 N_2 + \lambda_1 N_1 \quad (5)$$

$$\frac{dN_3}{dt} = C_3 D_3 - \lambda_3 N_3 + \lambda_2 N_2 \quad (6)$$

$$\frac{dN_4}{dt} = C_4 D_4 - \lambda_4 N_4 + \lambda_3 N_3 \quad (7)$$

Decay

$$\frac{dN_1}{dt} = -\lambda_1 N_1 \quad (8)$$

$$\frac{dN_2}{dt} = -\lambda_2 N_2 + \lambda_1 N_1 \quad (9)$$

$$\frac{dN_3}{dt} = -\lambda_3 N_3 + \lambda_2 N_2 \quad (10)$$

$$\frac{dN_4}{dt} = -\lambda_4 N_4 + \lambda_3 N_3 \quad (11)$$

Energy Spectra

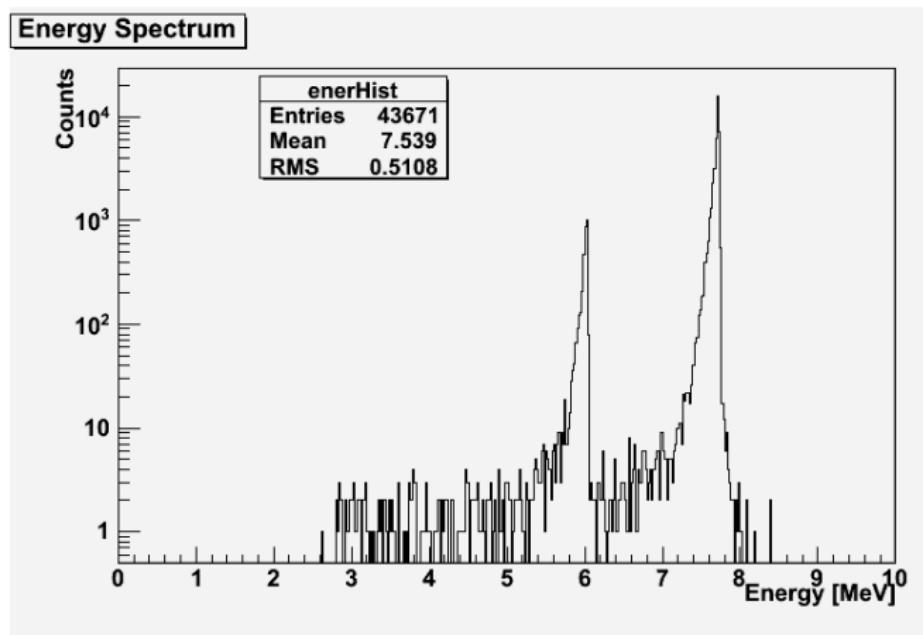


Figure: ^{218}Po around 6 MeV, ^{214}Po around 7.8 MeV

Decay Spectrum

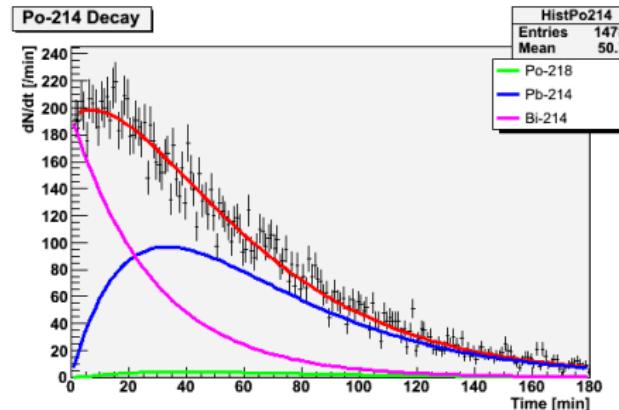
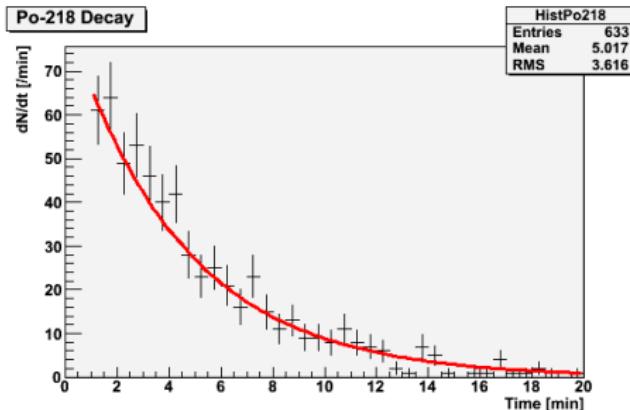
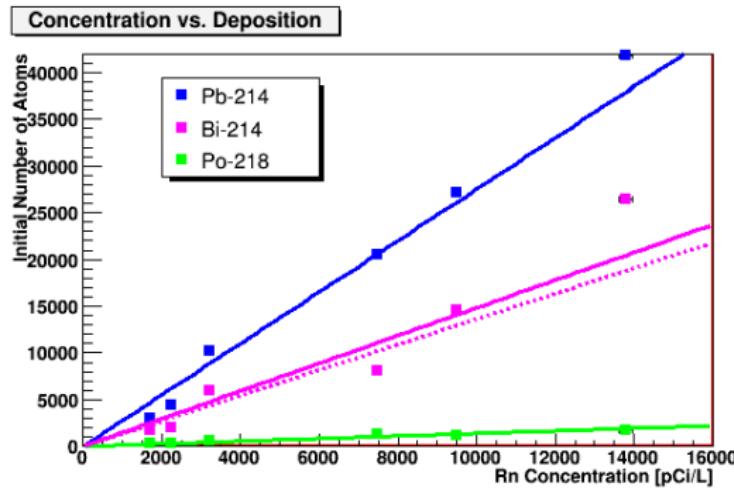


Figure: RH= 2%, Temp= 27.1 °C, Concentration= 15500 ± 108 pCi/L

$$[{}^{218}\text{Po}] = 1496.46 \pm 62.22 \text{ atoms}, [{}^{214}\text{Pb}] = 35770.7 \pm 469.65 \text{ atoms},$$
$$[{}^{214}\text{Bi}] = 22599.5 \pm 442.39 \text{ atoms}$$

Results

Deposition Rates for Acrylic



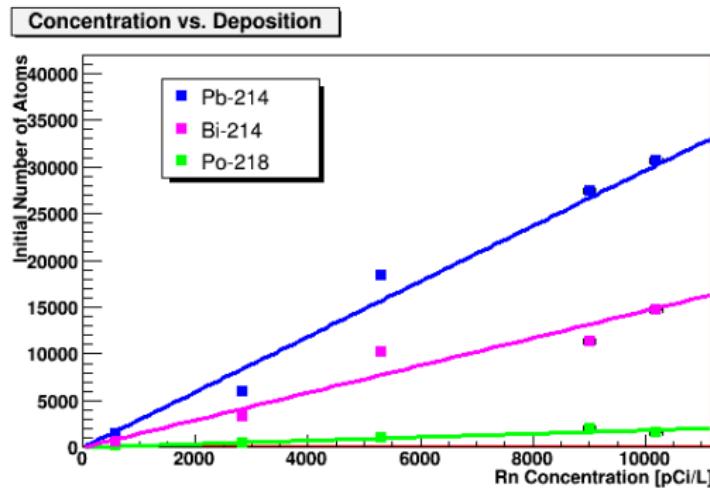
$$D_1 = 3.12 \times 10^{-2} \pm 7.78 \times 10^{-4} \text{ L/pCi/min/cm}^2$$

$$D_2 = 6.01 \times 10^{-2} \pm 8.85 \times 10^{-4} \text{ L/pCi/min/cm}^2$$

$$D_3 = 0 \pm 8.75 \times 10^{-4} \text{ L/pCi/min/cm}^2$$

Results

Deposition Rates for SUVT Acrylic



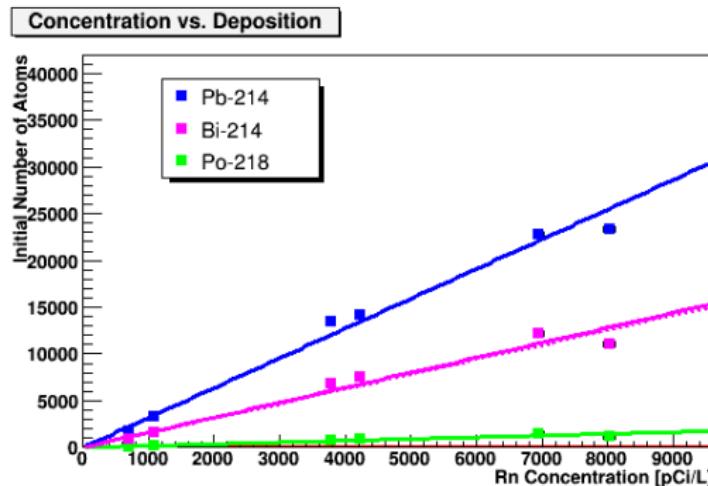
$$D_1 = 4.12 \times 10^{-2} \pm 1.10 \times 10^{-3} \text{ L/pCi/min/cm}^2$$

$$D_2 = 5.74 \times 10^{-2} \pm 1.12 \times 10^{-3} \text{ L/pCi/min/cm}^2$$

$$D_3 = 0 \pm 2.92 \times 10^{-4} \text{ L/pCi/min/cm}^2$$

Results

Deposition Rates for Copper



$$D_1 = 4.08 \times 10^{-2} \pm 1.18 \times 10^{-3} \text{ L/pCi/min/cm}^2$$

$$D_2 = 6.51 \times 10^{-2} \pm 1.23 \times 10^{-3} \text{ L/pCi/min/cm}^2$$

$$D_3 = 0 \pm 1.21 \times 10^{-3} \text{ L/pCi/min/cm}^2$$

Summary

	D_1 (L/pCi/min/cm ²)	D_2 (L/pCi/min/cm ²)	D_3 (L/pCi/min/cm ²)
Acrylic	$3.12 \times 10^{-2} \pm 7.78 \times 10^{-4}$	$6.01 \times 10^{-2} \pm 8.85 \times 10^{-4}$	$0 \pm 8.75 \times 10^{-4}$
SUVT Acrylic	$4.12 \times 10^{-2} \pm 1.10 \times 10^{-3}$	$5.74 \times 10^{-2} \pm 1.12 \times 10^{-3}$	$0 \pm 2.92 \times 10^{-4}$
Copper	$4.08 \times 10^{-2} \pm 1.18 \times 10^{-3}$	$6.51 \times 10^{-2} \pm 1.23 \times 10^{-3}$	$0 \pm 1.21 \times 10^{-3}$

- The deposition rates per concentration per unit area are as shown above
- Deposition rate varies linearly with radon concentration
- Sample material has little to no effect on deposition rate
- Deposition rate is insensitive to small fluctuations in temperature and particle count in deposition chamber

Continuing

- Repeat experiment at several varying
 - Temperatures
 - Humidity levels
 - Particle count
 - etc.
- Compare results using acrylic samples to those using copper samples

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