A Three-Step Kinetic Mechanism for Peptide Binding to MHC Class II Proteins†

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ABSTRACT: Peptide binding reactions of class II MHC proteins exhibit unusual kinetics, with extremely slow apparent rate constants for the overall association (<100 M⁻¹ s⁻¹) and dissociation (<10⁻⁵ s⁻¹) processes. Various linear and branched pathways have been proposed to account for these data. Using fluorescence resonance energy transfer between tryptophan residues in the MHC peptide binding site and aminocoumarin-labeled peptides, we measured real-time kinetics of peptide binding to empty class II MHC proteins. Our experiments identified an obligate intermediate in the binding reaction. The observed kinetics were consistent with a binding mechanism that involves an initial bimolecular binding step followed by a slow unimolecular conformational change. The same mechanism is observed for different peptide antigens. In addition, we noted a reversible inactivation of the empty MHC protein that competes with productive binding. The implications of this kinetic mechanism for intracellular antigen presentation pathways are discussed.

Proteins encoded by the major histocompatibility complex (MHC) gene locus bind peptide antigens and display them at the cell surface for inspection by the immune system as part of the mechanism by which foreign material in the body is recognized and removed (1). Class II MHC proteins generally are found on specialized immune system cells such as B cells, macrophages, and dendritic cells, but they can be expressed by most cell types in response to inflammation or infection (2). Newly synthesized class II MHC α- and β-glycoprotein subunits associate with a chaperonin-like invariant chain protein, which places an extended loop in the class II peptide binding site and directs transport to an endosomal compartment (3). Endosomal proteins cleave the invariant chain, and the bound fragment is exchanged for peptides generated from cell-surface and endocytosed proteins, in a poorly characterized process catalyzed by the peptide-exchange factor HLA-DM (4). MHC–peptide complexes then are transported to the cell surface for inspection by T-cell receptors on CD4⁺ T lymphocytes. Crystal structures have been determined for several human and murine class II MHC proteins in complex with defined peptides (5–13). In each case, the peptide was bound in a polyprolinalike II-like conformation, with several side chains projecting into specificity-determining pockets within the overall peptide binding groove, and with many additional contacts between the MHC proteins and the main chain of the bound peptide.

Initially, in vitro kinetic measurements of peptide binding to purified class II MHC proteins were interpreted in terms of the simple bimolecular reaction shown in Scheme 1 (14): Physical and chemical characterization of purified class II MHC revealed that they carried complex mixtures of tightly bound endogenous peptides (15–19). The stoichiometry of binding for peptide added to these preparations was quite low. Many of the natural peptide ligands had extremely long half-lives, often on the order of days, and could not be easily removed from the preparations without denaturing the MHC protein. Thus, for peptide binding in vitro to class II MHC proteins isolated from their native source, the predominant reaction is peptide exchange. Reactions of this type have been investigated in detail for several class II MHC proteins from humans and mice (20–22), but detailed kinetic analysis in this system is complicated by the exchange reaction and by heterogeneity in the mixture of endogenous peptides (23).

Production of empty, peptide-free class II MHC proteins simplified analysis of the binding reaction (24–28). Truncated class II MHC αβ subunits comprising the soluble extracellular domain could be produced in recombinant insect cells, which assemble the αβ heterodimer but do not load it with peptides (25). Recombinant HLA-DR1, a common human class II MHC protein, can be purified from such an expression system essentially free of endogenous peptides, and can be loaded to high efficiency with added peptides. For class II MHC proteins isolated from their native sources, the apparent forward rate constants determined in different laboratories varied from 1 to 100 M⁻¹ s⁻¹, with the slow kinetics attributed to the requirement for dissociation of endogenous peptide (14, 29, 30). The kinetics of peptide binding to the recombinant empty proteins were ~10-fold faster than for the native proteins, consistent with removal.
of the slow dissociation step from the overall reaction pathway (25, 31). Even for the empty protein, however, the apparent overall forward rate constant was still slow compared to the rates expected for a simple bimolecular reaction (25). This slow rate suggested a mechanism more elaborate than a simple bimolecular reaction (31—36). The observation that peptide dissociation kinetics and electrophoretic mobility of freshly formed complexes varied with incubation time suggested the presence of multiple peptide-bound species with different properties (31), consistent with either an on-pathway intermediate as shown in Scheme 2 or a nonproductive off-pathway reaction as shown in Scheme 3. These two possible pathways lead to different physical mechanisms for the binding reactions. In the mechanism shown in Scheme 2, an initial rapidly formed complex rearranges slowly to form a stable complex through conformational changes. In the mechanism shown in Scheme 3, most complexes form in the stable conformation. This latter mechanism might occur, for example, if some fraction of peptides (or MHC proteins) in solution are not in the proper conformation for classical binding, but are able to form alternative complexes with varying stability (31). Finally, another pathway has been proposed recently, based on a careful analysis of peptide exchange reactions, in which the MHC protein is present in active and inactive states (37, 38), as in Scheme 4. A similar pathway with alternate peptide states is possible in principle but unlikely given the unfolded structure expected for short peptides in solution.

To differentiate among potential mechanisms such as those shown in Schemes 2, 3, and 4, we developed a spectroscopic method for determination of complex formation based on fluorescence resonance energy transfer (FRET) between tryptophan residues in the MHC peptide binding site and a fluorescent reporter group attached to the peptide. Previous methods for analysis of the binding reaction required separation of reactants and products for analysis, preventing characterization of the early steps in the reaction. Using the fluorescence assay, we observed the binding behavior for two different peptides. The binding reaction can be described completely by a pathway incorporating elements of Schemes 2 and 4. In addition, we derived the relevant kinetic parameters for the binding reactions. This pathway has several implications for the mechanism of peptide binding within a cell.

**EXPERIMENTAL PROCEDURES**

**Peptides.** Peptide variants of CLIP (Ac-VSRMRMATP-L ôMQ, where ô is 2,4-diaminobutyric acid), derived from the class II-associated invariant chain (17), HA (Ac-PRFV-KQNTLRLAT), derived from influenza virus hemagglutinin (14), and MA (Ac-SGPLKAEIAQRLE), derived from influenza virus matrix protein (39—41), were synthesized by solid-phase Fmoc chemistry, deprotected, purified by reverse-phase HPLC, and analyzed by matrix-assisted laser desorption mass spectrometry and quantitative amino acid analysis to confirm the absence of incomplete synthetic or deprotection products. Underscored side chains indicate positions of unique amino groups used for introduction of fluorescent labels. Native peptide sequences were modified by K to R, and K to ô substitutions to facilitate single site labeling, at positions oriented away from the MHC protein, and were expected to have minimal effects on the MHC—peptide interaction based on structural modeling (9, 13). A Y-to-F change was made in the HA peptide to prevent intramolecular FRET. For fluorescent labeling, peptides were reacted overnight at room temperature in the dark with a 2-fold molar excess of 7-amino-4-methylcoumarin-3-acetic acid succinimide ester (AMCA-SE, Boehringer Mannheim) in 80% DMF or 100% DMSO, in the presence of 3 equiv of diisopropylaminoethanol (Sigma). Modified peptides were again purified and analyzed as above. AMCA-peptide concentrations were measured by the absorbance at 346 nm ($\epsilon_{346\text{ nm}} = 19,000 \text{ M}^{-1} \text{ cm}^{-1}$).

**HLA-DR1 and Peptide Complexes.** Soluble extracellular domains of HLA-DR1 were produced by baculovirus-directed secretion of recombinant protein from Sf9 insect cells with subsequent immunoaffinity purification, as described (25). Alternately, monomeric HLA-DR1 was produced by expression of α and β subunits in E. coli inclusion bodies with folding in vitro (28), followed by immunoaffinity purification to isolate properly folded HLA-DR1 and subsequent size-exclusion chromatography (Phenomenex HPSEC 3000) to isolate monomer (42). HLA-DR1 produced in Sf9 cells carries three N-linked high-mannose glycans, and C-terminal connecting peptide extensions to both subunits that are not present in HLA-DR1 produced in E. coli. Peptide-free (“empty”) HLA-DR1 preparations were concentrated and exchanged into PBS (7 mM Na+K+ phosphate, 135 mM NaCl, 0.02% sodium azide, pH 7.2) by centrifugal ultrafiltration (Amicon). The HLA-DR1 concentration was measured by the absorbance at 280 nm ($\epsilon_{280\text{ nm}} = 56,340 \text{ M}^{-1} \text{ cm}^{-1}$ for Sf9 DR1 and 54,375 M$^{-1}$ cm$^{-1}$ for E. coli DR1).

For preparation of DR1—peptide complexes, purified empty protein (10 $\mu$M) was incubated with a 2—3-fold molar excess of AMCA-peptide in PBST (PBS + 0.01% Tween-20) with protease inhibitors (0.1 mM iodoacetamide, 5 mM EDTA, 0.1 mg/mL PMSF) at 37 °C in the dark for 72 h, with subsequent isolation of peptide complexes by size-exclusion chromatography. Concentration and fractional peptide loading were determined by the absorbance at 280 nm using the appropriate $\epsilon_{280\text{ nm}}$ values for DR1 and AMCA (5300 M$^{-1}$ cm$^{-1}$) and $\epsilon_{346\text{ nm}}$ for AMCA. From the fluorescence emission intensities of free and bound peptides at equimolar concentrations, the AMCA-peptide extinction coefficient at 346 nm did not appear to change significantly after binding to DR1. Peptide binding capacities of DR1 preparations were measured by 12.5% SDS—PAGE after incubation as above, using the ability of DR1—peptide complexes to resist SDS-induced chain dissociation at room
temperature to quantitate the amount of bound complex formed (25, 28). Purified DR1–peptide complexes were kept in the dark at 4 °C.

To establish that the peptide labeling procedure did not substantially alter the binding affinity, apparent dissociation constants, \( K_{D,app} \), were measured by a competition assay with biotinylated HA peptide (bHA) and a sandwich ELISA using the LB3.1 antibody and streptavidin–europin as described previously (42, 43). For kinetically complex reactions such as those expected here, \( K_{D,app} \) determined by competition assay can vary depending on the assay conditions, and the relationship of \( K_{D,app} \) to the true equilibrium dissociation constant may not be straightforward (44). Peptides (10^{-10}–10^{-4} M) were incubated with 0.5 nM bHA and 0.5 nM empty Sf9 DR1 in PBST with 0.3% BSA and protease inhibitors at 37 °C for 3 days. An ELISA was used to detect the DR1–bHA complex formed in this reaction, and the IC_{50} values were converted to an apparent \( K_D \) using the equation: 

\[
K_{D,app} = \frac{IC_{50}}{1 + ([bHA]/K_{D,app, bHA])}.
\]

The apparent \( K_D \) of bHA, 14 nM, was determined in a direct binding experiment by incubating 0.5 nM empty DR1 and 2-fold dilutions of bHA (10^{-11}–10^{-5} M) under the conditions described above, using the ELISA to measure the DR1–bHA complex, and fitting the data to a quadratic equation describing a two-component binding reaction.

**Fluorescence Spectra and Standard Curves.** Fluorescence emission and excitation spectra were obtained using a SPEX Fluoromax-2 fluorescence spectrophotometer with slit widths of 1.0 nm (excitation) and 10.0 nm (emission). MHC-to-peptide FRET was measured using 285 nm excitation and 447 nm emission, which correspond to peaks in the protein fluorescence excitation and AMCA emission spectra, respectively. To convert fluorescence intensities to MHC–peptide complex concentrations, standard conversion factors were obtained from fluorescence spectra of purified DR1, purified AMCA-labeled peptides, and purified MHC–peptide complexes at various concentrations (5–500 nM) in PBST. The concentration dependence of fluorescence intensity was used to calculate the conversion factors: \( \gamma = 7900 \) cps/nM DR1–HA complex or \( \gamma = 7500 \) cps/nM DR1–CLIP complex and \( \delta = 1000 \) cps/nM for both free HA and CLIP peptide. For determination of the concentration of MHC–peptide complex (C) in a mixture with total peptide (P), the total fluorescence intensity (F) was measured and converted to concentration by the formula: 

\[
C = (F - \delta P)(\gamma - \delta).
\]

**Kinetic Measurements.** For measurement of dissociation kinetics, purified MHC–peptide complexes (30 nM) were incubated with a 100-fold excess of unlabeled peptide in PBST at 37 °C in the dark. At various times, samples were removed and immediately analyzed by size-exclusion chromatography in PBS (Phenomenex HPSEC3000), with the amount of complex determined using a Waters model 420-C fluorescence detector with 360 nm excitation and 425 nm long-pass emission filters. At the same time, another sample was removed, and fluorescence spectra were obtained as above for measurement of the complex by the FRET assay. Fluorescence values from size-exclusion chromatography were normalized by UV absorbance to control for variability of injection volume, and those from FRET spectra were normalized using the Raman spectra of a sample of deionized, distilled H_2O (\( \lambda_{ex} = 350 \) nm) to correct for day-to-day differences in lamp intensity.

For measurement of association kinetics as initial rates of reaction, solutions of DR1 or peptide were prepared in PBST at twice the desired final concentration (50 nM–10 μM) and were warmed to 37 °C. To start the reaction, equal volumes of protein solution and peptide solution were mixed together in a 5 mm path length cuvette, and the increase in the FRET signal (447 nm emission intensity using 285 nm excitation) was measured at various times, typically at 1 min intervals for 30 min. A constant concentration of DR1 (500 nM) was incubated with varying concentrations of peptide, or a constant concentration of peptide (100 nM) was incubated with varying concentrations of DR1. Initial rates of association \( (k_{obs}) \) were determined from the slope of a linear fit to the early (linear) part of each association curve. For comparison of these slopes, association data are presented as changes in the concentration of complex relative to that at time zero as extrapolated from a linear fit to the initial data points. This treatment removes the contribution of any burst phase of complex formation. For protein titrations, the burst phase amplitude was determined by subtraction of the prereaction protein and peptide fluorescence from the extrapolated time zero value. Increase in fluorescence was converted into ΔnM using the conversion factors described above, under the assumption of similar fluorescence intensities for the intermediate MHCpep’ and final MHCpep complex. We note that MHCpep’ and MHCpep need not have identical fluorescence intensities. For peptide titrations, the prereaction peptide fluorescence was large compared to the fluorescence changes in the initial phase of reaction, and the burst phase could not be determined accurately.

For determination of kinetic parameters \( k_2 \) and \( K_1 = k_{-1}/k_1 \), the initial rates of association \( (k_{obs}) \) were plotted against the concentration of either DR1 or peptide. These plots were fit via nonlinear least squares to a quadratic equation describing the rate of formation of MHC–peptide complex in a linear two-step reaction (Scheme 2), where the first step is in rapid equilibrium and the second step does not back-react under initial rate conditions. This equation is

\[
rate = k_1 \{(MHC_T + pep_T + K_i) \pm \sqrt{MHC_T^2 + 2K_i + 2pep_T} K_i + 2MHC_T K_i - 2MHC_T pep_T}/2 \}
\]

where MHC_T and pep_T are the total concentrations at time zero, \( K_i \) is the apparent equilibrium constant for the first step, and \( k_2 \) is the forward rate constant for the second step. The apparent active MHC fraction was determined by comparison of the best-fit MHC_T with the actual MHC concentration used in the experiment.

For measurements of the overall reaction progress, fluorescence changes were measured over several days for a reaction mixture containing 100 nM DR1 and 100 nM peptide at 37 °C. Fluorescence values from the progress curves were converted to concentration of MHC–peptide complex as described above and fit to single or double exponentials.

For measurement of the temperature dependence of the reaction, initial rate experiments were performed with 500 nM Sf9 DR1 and 50 or 300 nM HA peptide at 25, 30, 35, 40, and 45 °C. Initial rate versus temperature data were analyzed by a modified Arrhenius plot: In \( r = -E/RT + \ln A \), where \( r \) is the reaction rate (rather than the rate
constant), $E_A$ is the activation energy of the reaction, and $A$ is the frequency factor, using a least-squares fit.

For stopped-flow measurements, fluorescence data from a reaction between 1 μM DR1 and 100 nM HA peptide at 37 °C were collected on a logarithmic time base using a Hi-Tech SF-61 DX2 double mixing stopped-flow system with 285 nm excitation and detection at 425 nm. The dead time of this instrument is ~2 ms. The data were fitted to a single exponential using a nonlinear least-squares fit.

RESULTS

Fluorescence Energy Transfer Assay. To measure the kinetics of peptide binding to the human class II MHC protein HLA-DR1, we developed a fluorescence energy transfer method that allows assay of the binding reaction in situ and in real time. Previous studies of peptide binding to MHC proteins have been complicated by protocols that disturb the binding reaction to separate bound from free peptide for analysis, such as gel filtration or electrophoresis (37–38). The separation times for these methods can be long compared to the processes of interest, particularly for short-lived complexes or intermediate states. To measure peptide binding in real time, we used fluorescence energy transfer between MHC tryptophan residues and a fluorescent reporter group attached to the peptides. There are 10 tryptophan residues in DR1, with 4 in the peptide binding site and 6 in the lower immunoglobulin domains (5, 13). Of the residues in the binding site, two conserved tryptophans directly contact the bound peptide and two are within 15 Å from a peptide side chain. These are all well within the characteristic Förster transfer distance of ~30 Å expected for tryptophan and commonly used fluorescent probes (45, 46). After testing several probes for high quantum efficiency of transfer, maximum overlap between tryptophan emission and probe excitation, and minimum overlap between tryptophan excitation and probe excitation, we chose 7-amino-4-methylcoumarin-3-acetate (AMCA) as a fluorescent probe. The fluorophore was attached to the peptide ligands through a 2-acetamido ester linkage to a peptide lysine or diaminobutyric acid residue.

We used peptides based on the well-characterized DR1 ligands CLIP (Figure 1a) [from the class II-associated chaperonin invariant chain (17)] and HA (Figure 1b) and MAT (Figure 1c) [derived from influenza virus hemagglutinin and matrix proteins, respectively (39–41)]. Crystal structures are available for HA complexed with DR1 (13), and for CLIP complexed with the closely related allele HLA-DR3 (9). For the complex with MAT peptide, no structure is available, but the peptide register and orientation appear to be the same as observed in the DR1–HA and DR3–CLIP complexes, based on studies of the effects of peptide substitution on the binding affinity (47–49). To minimize any interaction of the AMCA probe with the DR1–peptide binding reaction, we attached the probe to side chains at positions oriented away from the binding site, at position P2 (MAT), P3 (HA), and P8 (CLIP), numbered relative to the large hydrophobic side chain at P1 that binds into MHC binding site pocket 1 (13). Introduction of the AMCA label did not affect the apparent affinity of peptide binding (Table 1). Additionally, Lys-to-Arg changes were introduced in MAT and HA to simplify labeling, and Tyr-to-Phe in HA to reduce intrapeptide energy transfer. These changes are not expected to substantially alter the peptide binding affinity (49, 50). For production of MHC–peptide complexes, soluble peptide-free DR1 was produced in baculovirus-infected insect cells as described (25), and loaded with AMCA-labeled peptides (see Experimental Procedures). In complex with DR1, each of the peptides exhibited increased fluorescence intensity at 400–500 nm with excitation at 285 nm, indicative of strong fluorescence resonance energy transfer (FRET) between aromatic amino acid residues in the MHC protein and the peptide AMCA (Figure 1). AMCA-MAT exhibited on binding to DR1 small shifts in both excitation and emission maxima and in probe quantum yield (Figure 1c, Table 1), indicating a physical interaction between
the AMCA probe and the MHC site. Because this interaction might alter the peptide—MHC binding reaction, this peptide was not investigated further. AMCA-HA and AMCA-CLIP were used to probe peptide binding kinetics.

To test the utility of the substantial difference in the fluorescence energy transfer for the bound versus free peptide in an assay of MHC peptide-binding kinetics, we measured dissociation kinetics for the DR1–CLIP complex (Figure 2a). The 447 nm emission band, representing energy transfer from DR1 to the bound peptide, decreases with increasing time as the peptide dissociates. The 330 nm emission band, representing emission from aromatic groups in uncomplexed DR1, increases concomitantly. The kinetics of these changes correlate with peptide release as measured in a conventional gel filtration assay performed on the same samples in parallel (Figure 2b). For CLIP, dissociation at 37 °C proceeds with \( k_{\text{off}} = 1.0 \times 10^{-5} \text{ s}^{-1} (\tau = 28 \text{ h}) \), in agreement with previous work (51). We were able to use the same system to measure association kinetics. An association reaction (Figure 2c) exhibited single-exponential binding kinetics (Figure 2d), with \( k_{\text{obs}} = 5.8 \times 10^{-4} \text{ s}^{-1} (\tau = 28.7 \text{ min}) \), corresponding to an apparent forward rate constant of \( \sim 100 \text{ M}^{-1} \text{ s}^{-1} \). This value is in accord with some but not all published values for this or similar reactions (14, 25, 29–31). Thus, the changes in fluorescence energy transfer reliably reflect peptide binding and dissociation.

**Initial Rates of Reaction.** We used the peptide association assay to investigate in detail the mechanism of peptide binding. Kinetic mechanisms such as shown in Schemes 1–4 can be distinguished based on the predicted reaction order

<table>
<thead>
<tr>
<th>peptide</th>
<th>excitation ( \lambda_{\text{max}}^a ) (nm)</th>
<th>emission ( \lambda_{\text{max}}^b ) (nm)</th>
<th>( F_{\text{bound}}/F_{\text{free}}^c )</th>
<th>( K_{d,\text{app}}(\text{unlabeled}) ) (nm)</th>
<th>( K_{d,\text{app}}(\text{MCA-labeled}) ) (nm)</th>
</tr>
</thead>
<tbody>
<tr>
<td>HA</td>
<td>285, 350 (350)</td>
<td>447 (445)</td>
<td>7.9</td>
<td>1.1</td>
<td>1.4</td>
</tr>
<tr>
<td>CLIP</td>
<td>285, 350 (350)</td>
<td>447 (445)</td>
<td>7.7</td>
<td>1.3</td>
<td>3.6</td>
</tr>
<tr>
<td>MAT</td>
<td>285, 360 (350)</td>
<td>432 (445)</td>
<td>8.9</td>
<td>n/d</td>
<td>n/d</td>
</tr>
</tbody>
</table>

\( ^a \) Fluorescence excitation maxima for 445 nm emission. The shorter wavelength represents energy transfer from DR1, and the longer wavelength direct excitation of the AMCA fluorophore. Values in parentheses are for the uncomplexed peptide. \( ^b \) Fluorescence emission maxima using 285 nm excitation. Values in parentheses are for uncomplexed peptide. \( ^c \) Ratio of fluorescence intensities for bound versus free peptide, using 285 nm excitation and 445 (HA, CLIP) or 432 nm (MAT) emission. \( ^d \) Not determined.
For each of the reactants. For example, in a kinetic process described by Scheme 1, the rate of production of MHC–peptide complex will be first order with respect to both MHC and peptide concentrations. A process described by Scheme 2 will behave similarly at low reactant concentrations, when the first step is rate-limiting, but the rate will become independent of both MHC and peptide concentrations at high reactant concentrations, when the second, unimolecular step is rate-limiting. A process described by Scheme 3 will behave similarly to one described by Scheme 1, with production of MHC–peptide complex first order with respect to both reactants. If the rates of formation and decay of the off-pathway product MHCpep′ are distinct from those of MHCpep, and if both MHCpep and MHCpep′ species are detected, biphasic association curves will be observed, with both kinetic phases exhibiting first-order kinetics. A process described by Scheme 4 always will be first order in MHC concentration. The peptide order will depend on the extent to which the first reaction contributes to the observed kinetics. If the first reaction is rate-limiting, the reaction can become independent of peptide concentration; otherwise, the process again reduces to one described by Scheme 1 but with a reduced MHC concentration. To distinguish among these mechanisms, we measured the dependence of the initial rate of binding on the concentrations of MHC and peptide, for both HA and CLIP peptides.

Figure 3a,b shows the dependence of the initial rate of binding on the concentration of HA, under conditions where the reaction has proceeded to less than 5% completion. The rate increases with increasing peptide at moderate concentrations but saturates at high peptide concentration. Similar behavior was observed with the CLIP peptide (Figure 3c,d). The behavior at lower concentrations indicates the reaction is first order with respect to peptide concentration. At high concentrations, the behavior indicates a rate-determining step that does not involve peptide binding. This rate saturation is inconsistent with a simple one-step binding reaction (Scheme 1) or a reaction in which two forward first-order reactions compete (Scheme 3), but is consistent with more elaborate reactions such as Scheme 2 or Scheme 4. In a separate experiment, the concentration of peptide was fixed, and the concentration of DR1 varied (Figure 4a,c). The rate of binding saturated with increasing concentration of DR1 for both HA and CLIP peptides (Figure 4b,d). This indicates that the rate-limiting step at high concentration does not directly involve DR1. This behavior is consistent with Scheme 2, but inconsistent with Scheme 4. Thus, taken together, these results suggest a mechanism in which a bimolecular step involving both peptide and MHC predominates at low concentrations, and a second step directly involving neither peptide nor MHC predominates at high concentrations.

For a reaction described by Scheme 2, formation of the intermediate species MHCpep′ in principle should be detectable by the FRET assay, as the energy transfer efficiency is likely to be substantial for any bound species. We attempted...
to investigate formation of a such a species as an early step in the peptide binding process that might be obscured by the mixing time (~30 s) of the initial rate experiments shown in Figures 3 and 4. We used a stopped-flow apparatus to measure the fluorescence changes during the early phase of a binding experiment (Figure 5a). The increase in fluorescence signal with time (10 ms–250 s) for the reaction of 100 nM AMCA-HA and 1 μM DR1 could be described by a single-exponential process with \( k_{\text{app}} = 2.7 \times 10^{-4} \text{ s}^{-1} \), consistent with the initial rates of reaction observed in Figures 3 and 4. Thus, any rapid kinetic step occurring early in the reaction was not resolved by the apparatus. However, we did find evidence for a rapid kinetic step in the comparison of the fluorescence intensities of each curve extrapolated to time zero in the initial rate experiment, with the intensities corrected for the calculated fluorescence of the unreacted protein and peptide components. This difference corresponds to a burst phase of complex formation (see Experimental Procedures). In Figure 4b, the amplitude of this burst phase is shown (open circles) along with the initial rate of complex formation (filled circles). For a reaction described by Scheme 2 with a rapid first step, both the initial rate of product formation and the burst phase can be observed.
formation and the burst phase amplitude will be proportional to the concentration of intermediate MHC\textsubscript{pep} formed, essentially as observed.

We measured the initial rates of reaction at various temperatures, in an attempt to resolve the two kinetic processes observed above. However, Arrhenius plots (Figure 5b) were similar under reaction conditions dominated by a unimolecular step (300 nM peptide) or reflecting both processes (50 nM peptide), with apparent activation enthalpies $\Delta H = 20 \text{ or } 23 \text{ kcal/mol}$, and entropies $\Delta S = -21 \text{ and } -18 \text{ cal mol}^{-1} \text{ K}^{-1}$, respectively.

**Kinetic Parameters.** For the linear, two-step mechanism of Scheme 2, when the first step is rapid and the reaction is under initial rate conditions so that the back-reaction is minimal (Scheme 5), the kinetic behavior can be described by two parameters: a quasi-equilibrium constant $K_1 = k_{-1}/k_1$ and a bimolecular forward rate constant $k_2$. Under conditions where MHC is in large excess and its concentration is essentially constant during the reaction, the rate of complex formation can be described by a hyperbolic function (eq 2):

$$\text{rate} = \frac{k_2[\text{pep}][\text{MHC}]}{k_{-1}/k_1 + [\text{MHC}]}$$

At MHC concentrations that are low relative to $k_{-1}/k_1$, the equation predicts second-order reaction kinetics with rate $= (k_2k_1/k_{-1})[\text{MHC}][\text{pep}]$. As the MHC concentration increases, the $k_{-1}/k_1$ term ceases to contribute to the observed rate, which reaches a limiting value of $k_2[\text{pep}]$. Without the assumption of one component in large excess, the rate is described by a quadratic equation that takes into account reactant depletion (eq 1, see Experimental Procedures), and which predicts a similar dependence of reaction order on reactant concentration. Using the quadratic equation, values were determined for the effective equilibrium constant $k_f/k_1$ and the saturating apparent rate constant $k_2$, for the peptide and protein titrations in Figures 3 and 4, for both HA and CLIP peptides (Table 2).

Surprisingly, the values for the kinetic parameters $k_{-1}/k_1$ and $k_2$ were different when derived from experiments with varying protein concentration than when they were derived from experiments with varying peptide concentration. The apparent constants $k_{-1}/k_{\text{app}}$ as determined in the constant peptide, variable protein titration were 25-fold (HA) and 52-fold (CLIP) greater than those determined in the constant protein, variable peptide titration (Table 2). The apparent saturating rate constants $k_2$ determined in the constant protein, variable peptide titration were 10–15-fold smaller than the same parameters measured in the variable protein, constant peptide titration. An explanation for this behavior would be that only a fraction of total MHC protein participates in the kinetic processes contributing to the initial rate data. The effect of a proportion of inactive MHC would be to increase the concentration of MHC protein needed to reach the half-maximal initial rate, leading to an apparent $k_{-1}/k_1$ greater than the true value. However, the determination of $k_2$ would be unaffected, as the effect of the inactive fraction would be overcome at high concentrations of total MHC protein. For the peptide titration, any inactive MHC would have no effect on the peptide concentration needed to reach the half-maximal initial rate, and the true $k_{-1}/k_1$ would be observed. However, the observed saturating rate $k_2$ determined in the constant protein, variable peptide titration was 20–100% of the total MHC protein was able to participate in the initial rate processes, with the value varying somewhat between MHC preparations. Using these values for the active fraction, we were able to determine corrected values for $k_{-1}/k_{\text{app}}$ from the protein titration, and $k_2$ from the peptide titration. Under this model, the kinetic parameters derived from protein and peptide titrations were approximately equal, with $k_{-1}/k_1 \sim 30 \text{ nM}$ and $k_2 \sim 1 \times 10^{-4} \text{ s}^{-1}$, for both HA and CLIP peptides (Table 2).

**Overall Reaction Progress.** We performed further experiments to characterize the nature of the inactive MHC fraction probed by the initial rate experiments. Although only 2–10% of the total MHC protein contributed to the initial rate processes, a much larger fraction, typically 50–90% for recombinant empty DR1 from various sources (25, 28), is able to bind peptide during a typical peptide loading protocol. For example, >65% of an empty DR1 preparation from Sf9 insect cells was able to bind peptide after 3 days at 37 °C, as measured by a gel electrophoresis assay (Figure 6a). This suggests that the fraction of MHC protein that is inactive in the initial rate experiments can convert during a long...
incubation into an active form competent to bind peptide. To investigate processes occurring over these longer time scales, we followed HA peptide binding by the FRET assay for several days (Figure 6a). Two kinetic processes were observed, one corresponding to the process described by the initial rate data, with \( \tau = 5 \) h, and one much slower, with \( \tau = 43 \) h. The final extent of peptide binding observed in this assay, \(~60\%\), agrees with that determined in the gel assay. Extended preincubation under the reaction conditions but in the absence of peptide had no effect on the MHC conversion (not shown). Thus, as the active protein binds peptide during the course of the reaction, inactive protein is converted slowly into the active form to maintain the equilibrium.

Monomeric MHC Class II from E. coli. One possibility for the slow activation process might be disaggregation of aggregated MHC protein. Empty MHC class II proteins produced in insect cells have a distinct tendency to aggregate (25), and usually are isolated as a mixture of monomers, dimers, and higher order oligomers with a specific profile on size-exclusion chromatography (Figure 7, dashed line). Upon peptide binding, the oligomerization of the empty protein is reversed, and peptide complexes are monomeric (25). Thus, the slow kinetic process observed above could represent MHC disaggregation driven by removal of monomeric MHC from the equilibrium mixture of aggregation states. To investigate this possibility, we made use of empty MHC class II proteins prepared by folding together DR\( \alpha \) and DR\( \beta \) subunits produced individually in E. coli inclusion bodies (28). The folding procedure can be carried out using low concentrations of protein in glycerol at 4 °C, conditions which minimize aggregation of the empty protein, and can be used to prepare monomeric empty DR1 essentially free of aggregated material (Figure 7, solid line) and fully competent to bind peptide (Figure 6b). Monomeric empty DR1 prepared by refolding will oligomerize very slowly under the assay conditions, with a time constant of \( \tau \approx 10^7 \) s or \(~9\) days (not shown). We used monomeric DR1 from E. coli to investigate the effect of aggregation on the kinetic steps described above. The dependence of the initial rates of reaction in both MHC and peptide titration experiments was similar to that observed for DR1 produced in insect cells, with bimolecular behavior at low concentration and saturating behavior at high concentrations (Figure 8). The high concentration regime of the monomeric DR1 titration plot (Figure 8b) could not be investigated due to MHC aggregation under these conditions; nonetheless, the concentration dependence of the initial rate is clearly hyperbolic and not linear. Kinetic parameters \( k_1 \) and \( k_2 \) derived from the fit to eq 2 differed in the peptide and protein titration experiments, in the same manner as values for DR1 produced in insect cells (Table 2).

![Figure 6: Overall reaction profile for (a) DR1/Sf9 and (b) DR1/E. coli.](image)

![Figure 7: Monomeric DR1 produced in E. coli.](image)
the inactive MHC fraction cannot be identified with aggregated protein, but rather with an inactive conformer present in both monomeric and oligomeric forms that can slowly convert to an active species during extended peptide binding reactions.

**DISCUSSION**

We have developed a fluorescence assay that we have used to monitor in real time the DR₁–peptide binding reaction in situ. The initial rate of the peptide binding reaction is concentration-dependent at low reactant concentrations, suggesting a bimolecular reaction step. At high reactant concentration, the initial rate no longer shows a concentration dependence, suggesting that a rate-limiting unimolecular step follows the initial bimolecular reaction. This behavior explains the observed variation in forward rate constants for previous studies of MHC peptide binding: the apparent rate constant depends on the concentration regime used for the measurement. The kinetic parameters extracted from experiments that measured the dependence of the initial rate on DR₁ concentration differed from those extracted from experiments which measured the dependence on peptide concentration. This behavior was observed for both HA and CLIP peptides. The differences were consistent with a small, active fraction of the total DR₁ preparation participating in the initial phase of the reaction. When the concentration of DR₁ was corrected to reflect the active percentage of protein participating in the reaction, the kinetic parameters extracted from either type of experiment were approximately equal.

The progress of the reaction tracked over a time course of days indicated that most of the total DR₁ eventually bound peptide. The progress curves for peptide binding were biphasic, with the first phase corresponding to the initial rate data, and the second phase corresponding to another step with a very long time scale. It appeared from these data that although a large fraction of DR₁ is not active initially, it becomes so over time. Because a monomeric form of DR₁ shows similar kinetic behavior to aggregated DR₁, the mechanism of inactivation is unlikely to be aggregation, and instead appears to be a conformational change in the empty protein. Overall, these kinetic data can be described by the three-step mechanism shown in Figure 9. Peptide rapidly binds to the active conformer MHCₐ to form the transient intermediate complex MHCpep'. We were not able to resolve the formation of this species with stopped-flow experiments (~2 ms resolution). The intermediate complex slowly undergoes a conformational change to form the stable MHCpep species. As the active conformer MHCₐ becomes depleted, the inactive conformer MHC, slowly converts into the active form competent to bind peptide.

Various aspects of the proposed mechanism have been suggested by previous studies. It has been suggested that peptide binding to empty MHC proteins occurs with rapid formation of a loosely bound intermediate complex which then slowly converts to a compact, peptide-loaded, conformation (31, 33). Hydrodynamic studies showing that the radius of empty DR₁ is 10–20% larger than for the stable, peptide-loaded complex support the existence of an “open” empty state which may be able to loosely bind peptides until conversion to the stable “compact” peptide-loaded conformation (42). For a murine class II MHC molecule, short- and long-lived MHCpep conformers were found in a pH-dependent equilibrium (52). However, the peptide binding reactions of HLA-DR₁, unlike those of some other class II MHC proteins, have been observed to have no major pH dependence (38, 53) (R.V.J. and L.J.S., unpublished observations). The postulated conversion between active and inactive MHC conformers (Figure 9) is supported by other recent studies (37, 38) showing that empty MHC proteins are in a “peptide-receptive” state when freshly dissociated from a complex with peptide, which we believe corresponds to the “active” MHC in our mechanism. The existence of different conformations for empty DR₁ is also supported by the wider distribution of hydrodynamic radii observed for empty
relative to peptide-loaded DR1 (42). The three-step mechanism proposed above affords a way to bridge all of these observations.

Our results suggest a possible mechanism of action for HLA-DM, the catalytic peptide exchange factor required for efficient peptide loading in vivo (4). The catalytic activity of HLA-DM was originally identified as increasing the rate of peptide exchange (54–56). Subsequently, HLA-DM was shown to promote peptide binding to empty HLA-DR1 (51) and to stabilize HLA-DR against inactivation in the absence of peptide (57). All of these activities could be explained by a model in which HLA-DM binds one of the intermediates transiently formed between MHC, and MHCpep, thereby lowering the barrier for both the forward and reverse reactions, and shifting the equilibria from MHC, to (Figure 9). The observation that HLA-DM catalyzes peptide exchange to an extent directly proportional to the off-rate of the peptide from the MHC (58) is consistent with either or both MHC, and MHCpep′ as the intermediate species bound by HLA-DM. In an attractive correlation between the kinetic species observed here and conformational variants observed in previous hydrodynamic and spectroscopic studies (42), both MHC, and MHCpep′ would correspond to the “open” conformation, and convert to the “closed” forms MHC, and MHCpep in kinetic steps $k_1$ and $k_2$.

Differences in the kinetic behavior of the DR1–HA and DR1–CLIP complexes can be traced to the kinetic stability of the final MHCpep complex. HA is a viral peptide that represents an immunodominant epitope in the response to influenza virus. CLIP is a fragment of the MHC-associated chaperonin invariant chain (Ii) that transiently occupies the MHC binding site during intracellular trafficking. HA and CLIP show similar behavior in the initial rate experiments; both bind rapidly to form transient intermediates with similar $k_{-1}/k_1$, which slowly convert to final MHCpep complexes with similar saturating rates $k_2$ (Figure 9). However, our measurements of the off-rates of each peptide indicate that CLIP’s off-rate is 10-fold higher than that of HA (Table 2). The final MHCpep complex with HA is long-lived, and in the biological context is essentially irreversible. The equilibrium between intermediate MHCpep′ and stable MHCpep conformations favors the formation of MHCpep for such irreversible complexes. On the other hand, the MHCpep complex with CLIP is unstable and dissociates relatively rapidly, with the equilibrium shifted to favor partial formation of the intermediate MHCpep′. The biological role of the CLIP peptide is to stabilize MHC class II molecules during transport and then to be exchanged for immunologically relevant endosomal peptides (2, 59). It is interesting that CLIP has apparently evolved to stabilize the MHCpep′ intermediate species for facile HLA-DM-mediated exchange.

How do these kinetic processes fit into cellular antigen metabolism? The inactive species MHC most likely does not normally occur in the cell, presumably due to the effect of HLA-DM (60) and invariant chain Ii. Newly synthesized class II MHC molecules immediately associate with invariant chain, which maintains the equilibrium toward complex formation and away from the inactive form. After trafficking of the MHC–Ii complex to the endosome under the influence of Ii targeting signals (61) and proteolysis of Ii in the endosome to leave CLIP in the binding site, the relevant reaction step would become peptide exchange under endosomal peptide concentration (62). Thus, MHC molecules enter into the kinetic mechanism at the MHCpep stage, with peptide exchange through MHCpep′ and MHC, facilitated by HLA-DM (Figure 9). The inactive form MHC would not become populated in vivo under endosomal peptide concentrations. Recognition and disposal of any endosomal MHC that had formed from molecules that did not acquire peptide despite prolonged residence in the endosome could act as a safety valve in B cells to prevent inappropriate surface expression of empty molecules. The physical basis for this recognition which regulates intracellular MHC trafficking may depend on the conformational transformations described by the three-step kinetic model.

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