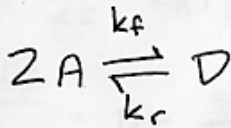


Lecture 1

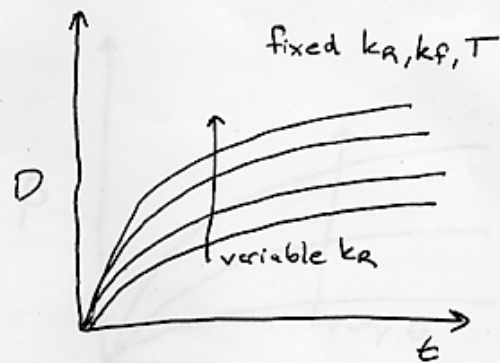
- the advantage of being dimensionless
- simplifying problems by comparing scales
- analyzing steady-state behavior
- ~~reputable~~ equations for gene regulation exhibiting self control
- a ~~control loop's self~~ : autoregulation

Dimerization

$$\frac{1}{2} \frac{dA}{dt} = -k_f A^2 + k_r D$$

$$\frac{dD}{dt} = -k_r D + k_f A^2$$

$$T = 2D + A$$



- varying k_A, k_f, T would require a book to convey system behavior
- use intuition to ~~reduce number~~ ^{combine} parameters

STEP 1: Identify all dimensions

$$A: c$$

$$T: c$$

$$D: c$$

$$k_A: 1/s$$

~~$$k_f: 1/c \cdot s$$~~

$$k_f: \frac{1}{c \cdot s}$$

$$t: s$$

$$\uparrow$$

parameters

BOOKKEEPING: variables

Example: Degradation

STEP 2: Introduce dimensionless variables

$$\alpha \equiv \frac{A}{T} \quad \beta \equiv \frac{P}{T} \quad \tau \equiv t \cdot k_R$$

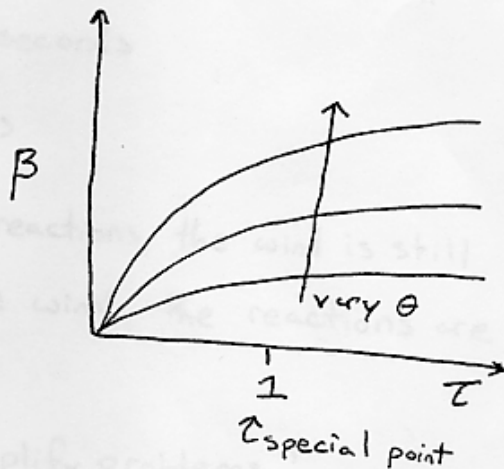
STEP 3: Substitute

$$\frac{dD}{dt} = k_R D - k_F A^2$$

$$k_R T \frac{d\beta}{d\tau} = k_R T \beta - k_F T^2 \alpha^2$$

$$\frac{d\beta}{d\tau} = \beta - \frac{k_F T}{k_R} \alpha^2$$

$$\boxed{\begin{aligned} \frac{d\beta}{d\tau} &= \beta - \theta \alpha^2 \\ \alpha + \beta &= 1 \end{aligned}}$$



We were lucky, dimerization is relatively easy

more formal definition

$$t_s \sim \frac{f_{\max} - f_{\min}}{\left| \frac{df}{dt} \right|_{\max}}$$

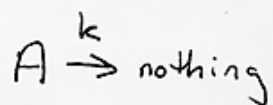
$1 = \tau_s = t_s k_R$
timescale of rxn:

$$t_s = \frac{1}{k_R}$$

if $k_R = 1/s \Rightarrow t_s$ is seconds

$k_R = 0.01/s \Rightarrow t_s$ is minutes

Example: Degradation



$$\frac{dA}{dt} = -kA \Rightarrow A(t) = A_0 e^{-kt}$$

BC: @ $t=0$, $A=A_0$

$$f_{\max} = A_0 \quad f_{\min} = 0 \quad \left| \frac{dA}{dt} \right|_{\max} = kA_0$$

$$t_s = \frac{A_0 - 0}{kA_0} = \frac{1}{k}$$

Mixing of Reactive Gasses:

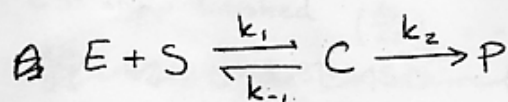
Ozone reactions: nanoseconds

Wind: weeks to months

- from the perspective of the reactions, the wind is still
- from the perspective of the wind, the reactions are always over

How can this be used to simplify problems?

Michaelis-Menton Kinetics



$$\frac{dS}{dt} = k_{-1}C - k_1ES$$

$$\frac{dC}{dt} = k_1ES - k_{-1}C - k_2C$$

velocity $v = \frac{dP}{dt} = k_2C$

$$E_0 = E + C$$

(4)

• common assumption: C is at quasi-steady-state

$$\frac{dC}{dt} = k_1 E_0 S - k_1 C S - k_{-1} C - k_2 C = 0$$

$$= k_1 E_0 S - (k_1 S + k_{-1} + k_2) C$$

$$C = \frac{k_1 E_0 S}{k_{-1} + k_2 + k_1 S} = \frac{E_0 S}{K_m + S} \quad K_m \equiv \frac{k_{-1} + k_2}{k_1}$$

$$v = \frac{dP}{dt} = \frac{k_2 E_0 S}{K_m + S} \quad \text{familiar form}$$

When is this assumption valid?

wind S

C "sees" constant S (S_0)

~~fast~~ ozone C

To S , C is always finished

~~C changes so fast~~

an argument of scale:

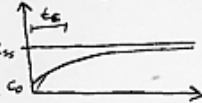
$$t_c \ll t_s$$

- constant $S = S_0$

$$\frac{dC}{dt} = k_1 E_0 S_0 - (k_1 S_0 + k_{-1} + k_2) C$$

$$\text{form of solution } C(t) = C_{ss} (1 - e^{-(k_1 S_0 + k_{-1} + k_2)t})$$

$$t_c \sim \frac{1}{k_1 S_0 + k_{-1} + k_2} = \frac{1}{k_1 (S_0 + K_m)} C_{ss}$$



- C is always finished $\left(\frac{dS}{dt}\right)_{qss} = -\left(\frac{dP}{dt}\right)_{qss}$ (mass conservation)

$$\frac{dS}{dt} = k_1 E_0 S + k_{-1} C - k_2 C - k_1 C S = 0$$

$$t_s = \frac{S_0 - 0}{\frac{k_2 E_0 S_0}{K_m + S_0}} = \frac{K_m + S_0}{k_2 E_0} \quad \& \quad \left|\frac{dS}{dt}\right|_{q, \max, qss} = \frac{k_2 E_0 S_0}{K_m + S_0}$$

Can now state conditions when g_{ss} is valid

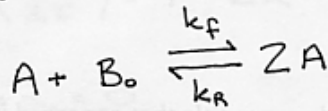
$$t_c \ll t_s$$

$$\frac{1}{k_1(s_0 + K_m)} \ll \frac{K_m + s_0}{k_2 E_0}$$

- t_s can be used to dedimensionalize the system

Autocatalysis

eg, Prions, 'RNA world'



Constant B_0

$$\frac{dA}{dt} = k_F A B_0 - k_R A^2$$

$$\alpha \equiv \frac{A k_R}{B_0 k_F}$$

$$Z \equiv t B_0 k_F$$

$$A = \frac{\alpha B_0 k_F}{k_R}$$

$$t = \frac{Z}{B_0 k_F}$$

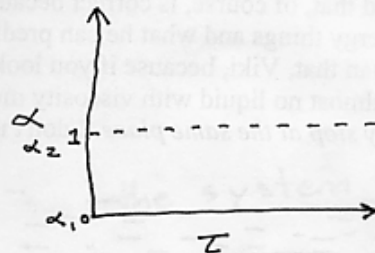
$$\left(\frac{B_0 k_F B_0 k_F}{k_R}\right) \frac{d\alpha}{dt} = \frac{k_F B_0 k_F B_0}{k_R} \alpha - \frac{k_R B_0^2 k_F^2}{k_R} \alpha^2$$

$$\boxed{\frac{d\alpha}{dt} = \alpha - \alpha^2}$$

at steady-state

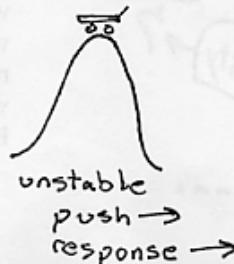
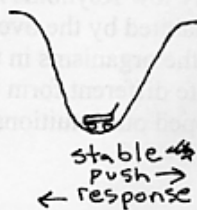
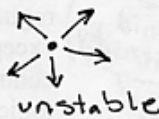
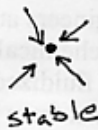
$$0 = \alpha - \alpha^2$$

$$\alpha_1 = 0, \alpha_2 = 1$$



• How do the solutions behave around the steady-states?

1-D problem has two options



Need to "mathematically nudge" the system at steady-state

"acceleration" of equations

$$\frac{\partial}{\partial \alpha} \left(\frac{d\alpha}{d\tau} \right) = 1 - 2\alpha$$

@ $\alpha = \alpha_1 = 0$

$$\left. \frac{\partial}{\partial \alpha} \left(\frac{d\alpha}{d\tau} \right) \right|_{\alpha=0} = 1 \quad \text{positive, } \underline{\text{unstable}} \text{ (system responds in the same direction it is pushed)}$$

@ $\alpha = \alpha_2 = 1$

$$\left. \frac{\partial}{\partial \alpha} \left(\frac{d\alpha}{d\tau} \right) \right|_{\alpha=1} = -1 \quad \text{negative, } \underline{\text{stable}} \text{ (system responds in the opposite direction it is pushed)}$$

timescale of the response

$$\tau_R \sim \frac{1}{\left. \frac{\partial}{\partial \alpha} \left(\frac{d\alpha}{d\tau} \right) \right|}$$

(the characteristic time for the system to recover)

