

Come up with an expression for the enthalpy and entropy departure functions for a gas that follows the Redlich-Kwong equation of state.

A. Entropy departure function

This problem is similar to Example 5.6 in Koretsky.

The Redlich-Kwong equation of state:

$$P = \frac{RT}{v-b} - \frac{a}{T^{1/2}v(v+b)}$$

As the Redlich-Kwong equation is explicit in pressure, it is convenient to choose T and v as the independent variables. Thus infinite volume is used as the limit of a ideal gas.

$$\Delta h_{T,v}^{dep} = h_{T,v} - h_{T,P}^{ideal} = (h_{T,v} - h_{T,v=}^{ideal}) - (h_{T,P}^{ideal} - h_{T,v=}^{ideal})$$

For an ideal gas, enthalpy is only a function of temperature. Thus

$$\Delta h_{T,v}^{dep} = h_{T,v} - h_{T,v=}^{ideal}$$

$$\Delta h_{T,v}^{dep} = \int_{v=}^v \frac{h}{v} \frac{dv}{T} \quad (1)$$

(1 point)

To find $\left(\frac{\partial h}{\partial v}\right)_T$, use the Equation 5.7 and Equation 5.18,

$$\left(\frac{\partial h}{\partial v}\right)_T = \frac{T}{v} \left(\frac{\partial s}{\partial v}\right)_T = T \left(\frac{\partial s}{\partial v}\right)_T + v \left(\frac{\partial P}{\partial v}\right)_T = T \left(\frac{P}{T}\right)_v + v \left(\frac{\partial P}{\partial v}\right)_T \quad (2)$$

(1 point)

To find $\left(\frac{\partial P}{\partial T}\right)_v$ and $\left(\frac{\partial P}{\partial v}\right)_T$, integrate the RK EOS,

$$\frac{P}{T} = \frac{R}{v} + \frac{a}{b + 2T^{3/2}v(v-b)} \quad (1 \text{ point})$$

$$\frac{+P}{-v} \bigg|_T = -\frac{RT}{(v-b)^2} + \frac{a(2v-b)}{T^{1/2}v^2(v-b)^2} \quad (1 \text{ point})$$

Substitute expressions for $\left[\frac{P}{T} \right]_v$ and $\left[\frac{P}{v} \right]_T$ into (2) and simplify

$$\frac{+h}{-v} \bigg|_T = \frac{RT}{v-b} + \frac{aT}{2T^{3/2}v(v-b)} - \frac{RTv}{(v-b)^2} + \frac{av(2v-b)}{T^{1/2}v^2(v-b)^2}$$

$$\frac{-h}{+v} \bigg|_T = \frac{RTb}{(v-b)^2} + \frac{5a}{2T^{1/2}(v-b)^2} + \frac{3ab}{2T^{1/2}v(v-b)^2}$$

Substitute the above expression into Equation (1).

$$\Delta h_{T,v}^{dep} = \int_{v=0}^v \left[\frac{-RTb}{(v-b)^2} + \frac{5a}{2T^{1/2}(v-b)^2} + \frac{3ab}{2T^{1/2}v(v-b)^2} \right] dv \quad (1 \text{ point})$$

Integrate the above expression using Mathematica or MATLAB.

$$\Delta h_{T,v}^{dep} = \frac{RTb}{v-b} - \frac{5a}{2T^{1/2}(v-b)} + \frac{3a \left(b + (b+v) (\ln v - \ln(v+b)) \right)}{2bT^{1/2}(v-b)} \quad (1/2 \text{ point})$$

B. Entropy departure function

$$\Delta s_{T,v}^{dep} = s_{T,v} - s_{T,v}^{ideal} = (s_{T,v} - s_{T,v}^{ideal}) - (s_{T,v}^{ideal} - s_{T,v}^{ideal})$$

Note that v^{ideal} is the volume of the gas at ideal condition. It is not equal to v .

$$\Delta s_{T,v}^{dep} = \int_{v}^v \frac{s}{v} dv - \int_{v}^{v^{ideal}} \frac{s}{v} dv \quad (3)$$

(1 point)

Using Maxwell Relations Equation 5.18

$$\left(\frac{\partial s}{\partial v} \right)_T = \frac{P}{T}$$

Thus Equation (3) becomes

$$\Delta s_{T,v}^{dep} = \int_{v}^v \frac{P}{T} dv - \int_{v}^{v^{ideal}} \frac{P}{T} dv \quad (4)$$

(1 point)

From earlier part, $\left(\frac{\partial P}{\partial T} \right)_v$ for real gas:

$$\frac{P}{T} = \frac{R}{v-b} + \frac{a}{2T^{3/2}v(v+b)}$$

For an ideal gas,

$$\frac{P}{T} = \frac{R}{v}$$

Substitute the above expressions into Equation (4) and integrate

$$\Delta s_{T,v}^{dep} = \int_{v}^v \frac{R}{v-b} + \frac{a}{2T^{3/2}v(v+b)} dv - \int_{v}^{v^{ideal}} \frac{R}{v} dv \quad (1 \text{ point})$$

To eliminate the problem of singularity in the second integral, add and

subtract $\int_{v=}^v \frac{R}{v} dv$

$$\Delta s_{T,v}^{dep} = \int_{v=}^v \frac{R}{v-b} + \frac{a}{2T^{3/2}v(v+b)} - \frac{R}{v} dv - \int_{v=}^{v^{ideal}} \frac{R}{v} dv \quad (1 \text{ point})$$

$$\Delta s_{T,v}^{dep} = R[\ln(v-b) - \ln v] + \frac{a(\ln v - \ln(v+b))}{2bT^{3/2}} - R \ln \frac{v^{ideal}}{v} \quad (1/2 \text{ point})$$

Note on the infinity limit:

$$\begin{aligned} \Delta s_{T,v}^{dep} &= \int_{v=}^v \frac{P}{T} dv - \int_{v=}^{v^{ideal}} \frac{P}{T} dv \\ &= \int_{v=}^v \frac{P}{T} dv - \int_{v=}^{v^{ideal}} \frac{R}{V} dv \end{aligned}$$

To eliminate the problem of singularity in the second integral, add and

subtract $\int_{v=}^v \frac{R}{v} dv$

$$\Delta s_{T,v}^{dep} = \int_{v=}^v \frac{P}{T} - \frac{R}{V} dv - \int_{v=}^{v^{ideal}} \frac{R}{V} dv$$

You can combine the two terms in the first integral

$$\Delta s_{T,v}^{dep} = \int_{v=}^v \left(\frac{P}{T} - \frac{RT}{V} \right) dv - \int_{v=}^{v^{ideal}} \frac{R}{V} dv$$

At the limit of $v \rightarrow \infty$, $P \rightarrow \frac{RT}{V}$. Thus the integral goes to zero at limit of $v \rightarrow \infty$.

This is why the first integral can have infinity as a limit but not the second integral.