# Symmetry Elements and Operations 

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## Outline

(1) Symmetry Elements

- Planes of Reflection, $\sigma$
- Axes of Rotation, $C_{n}$
- The Inversion Center, $i$
- Improper Axes of Rotation, $S_{n}$
- The Identity, $E$
(2) Symmetry Operations


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(2) Symmetry Operations


## A Molecule with Two Mirror Planes

The $\mathrm{H}_{2} \mathrm{O}$ Molecule

- The O and H atoms lie in the same plane
- The plane of the molecule is a mirror plane, $\sigma$
- The plane $\perp$ to the molecular plane is a second $\sigma$
- The molecular plane is taken as the $y z$ plane
- The mirrors are $\sigma_{v}(x z)$ and $\sigma_{v}^{\prime}(y z)$
- The subscript $v$ is for "vertical"


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## Some Other Molecules

Does the molecule have a mirror plane?

- Ammonia, $\mathrm{NH}_{3}$
- Sulfur tetrafluoride, $\mathrm{SF}_{4}$
- Dioxygen, $\mathrm{O}_{2}$
- White phosphorus, $P_{4}$
- Diborane, $\mathrm{B}_{2} \mathrm{H}_{6}$
- Myoglobin


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- The Identity, E
(2) Symmetry Operations


## Molecules with Axes of Rotation, $C_{n}$

These have $n$-fold axes of rotation

## What is $n$ ?

- Ammonia, $\mathrm{NH}_{3}$
- Cubane, (CH)8
- Water, $\mathrm{H}_{2} \mathrm{O}$
- Buckminsterfullerene, $C_{60}$
- Tick-Borne Encephalitis Virus


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## Molecules with Axes of Rotation, $C_{n}$

Some observations based upon symmetry

- Ammonia, $\mathrm{NH}_{3}$, has a single "higher-order" axis of rotation denoted $C_{3}$
- Molecules with a $C_{n}$ axis where $n \geq 3$ have degenerate electronic energy levels
- Molecules with a $C_{n}$ axis where $n \geq 3$ have degenerate vibrational energy levels
- Water, $\mathrm{H}_{2} \mathrm{O}$, has a single $C_{2}$ axis and thus has no degeneracies required by symmetry


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(2) Symmetry Operations


## Molecules with Centers of Inversion

These have identical atoms with inverted coordinates

## Is there an inversion center, i?

- Ethylene, $\mathrm{C}_{2} \mathrm{H}_{4}$
- Methane, $\mathrm{CH}_{4}$
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## Molecules with an Improper Axis of Rotation A Combination of Rotation Axis and Mirror Plane

Consider Pt(SH) ${ }_{4}{ }^{2-}$

- Pt-S-H bonds are bent
- H atoms are located alternately above and below the $\mathrm{PtS}_{4}$
plane
- There is not a $C_{4}$ axis
- There is not a mirror plane containing the four $S$ atoms
- There is an $S_{4}$ axis passing through Pt and $\perp$ to sulfur plane


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## The Identity Symmetry Element

This is present by default

- Rotation by $360^{\circ}$ about an arbitrary axis returns an equivalent configuration
- This axis is referred to as the identity symmetry element, $E$


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## Symmetry Operations

These are carried out with respect to symmetry elements

- A mirror plane, $\sigma$, generates a single reflection operation
- Two consecutive reflections with respect to a given $\sigma$ is equivalent to the identity operation
- $A C_{2}$ axis generates a single two-fold rotation operation
- $\mathrm{A} C_{3}$ axis generates two operations: rotation by $\frac{2 \pi}{3}$ and rotation by $\frac{4 \pi}{3}$
- The latter operations are called $C_{3}$ and $C_{3}^{2}$, respectively



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