5.03, Inorganic Chemistry Prof. Daniel G. Nocera Lecture 2 May 11: Ligand Field Theory

The ligand field problem is defined by the following Hamiltonian,



The Hamiltonian contains three types of terms arising from the electrons interacting with the positively charged nucleus of the atom (defined by $1e^-$ interactions, r_i), $2e^-$ interactions arising from e^--e^- repulsions (r_{ij}), the interaction of electrons with electron cloud of the ligands ($|R-r_i|$ where R is the position of the ligand),



The electronic part of the Hamiltonian is spectroscopically probed in two regimes, depending on the strength of the ligand field.

In the weak field, the 2e⁻ energies are greater than the 1e⁻ energies, i.e. $Q_{ij}\left(\frac{1}{r_{ij}}\right) >> V_{O_h}$

of which Δ_0 is a measure of V_{O_h} . So the strategy here is to first determine state symmetries followed by applications of ligand field.



In the **strong field**, the 1e⁻ energies (orbital energies) are first determined, followed by the perturbation of 2e⁻ terms... begin with configurations and see how they are perturbed by Q_{ij} .



The overall approach to determining the energy levels of a molecule is:

- ✤ determine free ion states
- ↔ determine how atomic states are split in a weak field limit (PE >> Δ_0)
- ↔ determine states arising from configurations in a strong field limit (PE << Δ_0)
- correlate between strong and weak fields (Tanabe-Sugano diagrams).

The Weak Field

The problem begins by considering a free ion (K_h symmetry). Different atomic configurations give rise to different **terms** or **energy levels**. The many e⁻ system is characterized by a total orbital and total spin angular momentum,

total orbital angular momentum, L

$$\begin{split} L &= \sum_{i}^{n} \ell_{i} \qquad L = 0, \ 1, \ 2, \ 3... \\ M_{L} &= \sum_{i}^{n} m_{\ell_{i}} \qquad M_{L} = L, \ L-1, ... \ 1-L, \ -L \end{split}$$

total spin angular momentum, S

$$S = \sum_{i}^{n} s_{i}$$
 $S = \frac{1}{2}, 1, \frac{3}{2}...$

$$M_{s} = \sum_{i}^{n} m_{s_{i}}$$
 $M_{s} = S, S-1 ... 1-S, -S$

A term is symbolized by



We will therefore need to

- 1) determine terms of a given configuration
- 2) determine the eigenfunction of these terms
- 3) calculate terms energies

As an illustrative example, consider the p^2 configuration,



A shorthand notation for this configuration is $(\hat{1} \ \hat{0})$, which is a **Slater determinant**

$$(\stackrel{+}{1} \stackrel{-}{0}) = \frac{1}{\sqrt{2}} \begin{vmatrix} (\stackrel{+}{1})^{1} & (\stackrel{+}{1})^{2} \\ (\stackrel{-}{0})^{1} & (\stackrel{-}{0})^{2} \end{vmatrix} = \frac{1}{\sqrt{2}} \begin{bmatrix} (\stackrel{+}{1}, \stackrel{-}{0}) - (\stackrel{-}{0}, \stackrel{+}{1}) \\ (\stackrel{+}{1}, \stackrel{-}{0}) - (\stackrel{-}{0}, \stackrel{+}{1}) \end{bmatrix}$$

All the possible configurations are summarized in the following table which defines the configurations in terms of M_L and M_S

M₅ M∟	1	0	-1
2		(1, 1)	
1	(1, 0)	$(\stackrel{+}{1},\stackrel{-}{0})(\stackrel{-}{1},\stackrel{+}{0})$	$(\bar{1},\bar{0})$
0	(1,-1)	$(\overset{+}{0}, \overset{-}{0})$ $(\overset{+}{1}, -\overset{-}{1})$ $(\overset{-}{1}, -\overset{+}{1})$	(1,-1)
-1	(⁺ 0,- ⁺ 1)	$(\overset{+}{0},-\overset{-}{1})(\overset{-}{0},-\overset{+}{1})$	(0,-1)
-2		(-1,-1)	

The table summarizing the number of constituent configurations in terms of M_L and M_S ,

M _s M∟	1	0	-1
2		1	
1	1	2	1
0	1	3	1
-1	1	2	1
-2		1	

Considering the definitions of L and S, begin with highest M_L wavefunction and eliminate $1e^-$ wavefunctions from +L to -L and +S to -S. The highest M_L is 2, $M_S = 0$

 $M_L = 2$, $M_S = 0$, which is a ¹D term

Removing the count of configurations arising from a 1D term (i.e., from M_L = 2, 1, 0, -1, -2 and M_S = 1, 0, -1), leaves

Ms M∟	1	0	-1
2			
1	1	1	1
0	1	2	1
-1	1	1	1
-2			

The highest M_L is 1 and highest M_S of this M_L value is also 1, thus giving rise to a ³P term

Again, eliminating states of the ³P term,

M₅ M∟	1	0	-1
2			
1			
0		1	
-1			
-2			

This leaves only one configuration of $M_L = 0$ and $M_S = 0$, which is a ¹S term

Thus the terms arising from a p^2 configuration are: ¹D, ³P, and ¹S. May predict the ground state from Hund's rule:

- 1) state with maximum M_s is lowest in energy
- 2) for states of the same spin multiplicity, the state with largest $M_{\mbox{\tiny L}}$ will be lowest in energy
- 3) note, Hund's rule does not address excited state ordering

Hund's rule identifies ³P as the ground state. Summarizing the energy (term) levels



this ordering must be determined from spectroscopy

How about the d ion states?

 d^1 ion

... is straightforward. There are 10 permutations of an electron in 5 d orbitals, s = $\frac{1}{2}$ or s = $-\frac{1}{2}$ in orbitals of m_l of 2 to -2



This gives $M_L = 2$, 1, 0, -1, -2 and $M_S = 1$, 0, -1, thus giving values of L = 2 and S = 1 or a ²D state. This is the only term for the d¹ configuration

 d^2 ion

... has many terms that result from the permutation of 2e⁻ in the orbital subshells derived from ℓ = 2,m_{ℓ} = 2, 1, 0, -1, -2

M₅ M∟	1	0	-1
4		(2,2)	
3	(2,1)	(⁺ 2, ⁻ 1) (⁻ 2, ⁺ 1)	$(\bar{2}, \bar{1})$
2	(2,0)	$(\overset{+}{2},\overset{-}{0})(\overset{+}{1},\overset{-}{1})(\overset{-}{2},\overset{+}{0})$	(2,0)
1	(⁺ 2,- ⁺ 1) (⁺ 1, ⁺ 0)	$(\overset{+}{2},-\overset{-}{1})(\overset{+}{1},\overset{-}{0})(\overset{-}{1},\overset{+}{0})(\overset{-}{2},-\overset{+}{1})$	(2,-1) (1,0)
0	(⁺ 2,- ⁺ 2) (⁺ 1,- ⁺ 1)	$(\stackrel{+}{2},-\stackrel{-}{2})(\stackrel{+}{1},-\stackrel{-}{1})(\stackrel{-}{2},-\stackrel{+}{2})$ $(\stackrel{-}{1},-\stackrel{+}{1})(\stackrel{-}{0},\stackrel{+}{0})$	(2,-2) (1,-1)

where the M_L terms are not shown for the convenience of space. Reducing the terms yields for a d² system: $^3F,\,^3P,\,^1G,\,^1D$ and 1S

The Russell-Saunders terms for all dⁿ free ions:



Now need to determine the states that arise from the application of an $O_{\rm h}$ field on the free ion.

The degeneracy of the wavefunction is removed upon application of the ligand field. The ligand field causes the terms to split owing to the change in symmetry from K_h to O_h . We can use the basis functions of the Oh character table to determine how the fee ion states split,

Taking the terms of a p^2 ion in a spherical field (K_h) to an O_h ligand field leads to the further splitting,



The Strong Field

The problem in the strong field begins with the electron configurations derived from the molecular orbital. As an example, consider the strong field configurations for two electrons in an octahedral ligand field. The two-electron occupancy of the $ML_6(O_h)$ MO gives rise to three configurations:

$(t_{2g})^2$	<	$(t_{2g})^1(e_g)^1$	<	(e _g) ²
ground state		1e ⁻ excitation		2e ⁻ excitation

Must now determine what states from d^2 weak field correlate to these configurations. Because the electrons interact differently with each other, when in different orbitals (different r_{ij}), for a given electronic configuration, many states can arise.

Let's begin with the 1e⁻ excited state, $(t_{2g})^1(e_g)^1$. The state symmetry takes the symmetry of the individually occupied orbitals, and the electrons can be singlet or triplet paired, ${}^1(\chi\cdot\chi')$ and ${}^3(\chi\cdot\chi')$



The spin can be S = 0 (singlet) or S = 1 (triplet), thus all potential states that arise from the one electron excitation of a d^2 ion in a O_h field are:

 ${}^{1}T_{1g} + {}^{1}T_{2g} + {}^{3}T_{1g} + {}^{3}T_{2g}$

The $(t_{2g})^2$ and $(e_g)^2$ configurations are more problematic because the electrons are degenerate. The direct product electrons in n degenerate orbitals of χ and χ' symmetry will give $\chi \cdot \chi'$ orbital symmetries.

$$(t_{2g})^2 \implies t_{2g} \times t_{2g} = A_{1g} + E_g + T_{2g} + T_{1g}$$

 $(e_g)^2 \implies e_g \times e_g = A_{1g} + A_{2g} + E_g$

But there is a problem, t singlet and triplet spin pairing cannot be assigned to each state because some of these states will violate the Pauli exclusion principle. In 5.04 a procedure will be derived that shows how to eliminate these Pauli violating states. Here the result will be given:

$$(t_{2g})^{2} \implies t_{2g} \times t_{2g} = {}^{1}A_{1g} + {}^{1}E_{g} + {}^{1}T_{2g} + {}^{3}T_{1g}$$
$$(e_{g})^{2} \implies e_{g} \times e_{g} = {}^{1}A_{1g} + {}^{3}A_{2g} + {}^{1}E_{g}$$

The states created in the weak field must be the same as the states obtained in the strong field since we are dealing with the d^2 problem, just from different perspectives. We see this is the case. A correlation diagram relates the weak and strong field configurations. The d^2 correlation diagram is:



Note that for d^2 (d^n) in O_h , the same diagram is obtained for d^8 (d^{10-n}) T_d . T_d symmetry does not require the g or u subscripts.